

DOCKET NO: 550P002Cont.2

PATENT

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In Re Application  
of:

William Kopaciewicz

Serial No.: Not yet assigned  
Group Art Unit: Not yet assigned  
Filing Date: Herewith by Express Mail  
Examiner: Not yet assigned

For: CAST MEMBRANE STRUCTURES FOR SAMPLE PREPARATION

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DATE OF DEPOSIT: September 1, 1999

BOX: ☒ Patent Application

☐ Provisional ☐ Design ☐ Sequence

Assistant Commissioner for Patents  
Washington, D.C. 20231

Sir:

PATENT APPLICATION TRANSMITTAL LETTER

Transmitted herewith for filing, please find:

☒ A Utility Patent Application.

If this is a continuing application, please check appropriate  
box:

☒ Continuation ☐ Divisional ☐ Continuation-in-part  
of prior application U.S. Serial No. 09/007,320 filed  
January 15, 1998

☐ A Design Patent Application.

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Including the following:

- ☐ Provisional Application Cover Sheet.
- ☒ New or Revised specification, including pages 32 & 33 containing:  
☐ Specification  
☒ Claims  
☐ Abstract
- ☐ A copy of earlier application Serial No. \_\_\_\_\_ Filed \_\_\_\_\_ to which no new matter has been added, TOGETHER WITH a copy of the executed oath or declaration for such earlier application and all drawings and appendices. Such earlier application is hereby incorporated into the present application by reference.
- ☒ A copy of executed oath or declaration for earlier application Serial No. 09/007,320 Filed January 15, 1998. The entire disclosure of such earlier application is considered as being part of the disclosure of the present application and is hereby incorporated into the present application by reference.
- ☒ Please enter the following amendment to the Specification under the Cross Reference to Related Applications section (or create such a section): "**This application is a Continuation of U.S. Serial No. 09/007,320 filed January 15, 1998**"
- ☒ Signed Statement attached deleting inventor(s) named in the prior application.
- ☐ Preliminary Amendment is enclosed.
- ☒ Nineteen (19) Sheet(s) of Drawings.
- ☐ Petition to Accept Photographic Drawings.  
☐ Petition Fee
- ☐ An ☐ Executed ☐ Unexecuted Declaration or Oath and Power of Attorney.
- ☐ An Associate Power of Attorney.
- ☐ An Executed Assignment of the Invention to:  
\_\_\_\_\_
- ☐ A Recordation Form Cover Sheet.
- ☐ Recording Fee - \$40.00.

DOCKET NO: 550P002Cont.2

☒ Priority is claimed under 35 U.S.C. §119 of U.S. Provisional application serial no: 60/038,909 filed February 26, 1997

A certified copy of the foregoing application for which priority is claimed:

\_\_\_\_\_ is enclosed.

\_\_\_\_\_ will be submitted in due course

\_\_\_\_\_ was filed in prior application Serial No. \_\_\_\_\_  
filed \_\_\_\_\_

\_\_\_\_\_ was transmitted by the International Bureau under International Application No. \_\_\_\_\_ filed \_\_\_\_\_

\_\_\_\_\_ An \_\_\_\_\_ Executed \_\_\_\_\_ Copy of Earlier Statement Claiming Small Entity Status under 37 C.F.R. 1.9 and 1.27

\_\_\_\_\_ is enclosed.

\_\_\_\_\_ has been filed in prior application Serial No. \_\_\_\_\_  
filed \_\_\_\_\_, said status is still  
proper and desired in present case.

\_\_\_\_\_ Diskette Containing DNA/Amino Acid Sequence Information.

\_\_\_\_\_ Statement to Support submission of DNA/Amino Acid Sequence Information.

\_\_\_\_\_ Letter of Reference to Computer Readable Form.

\_\_\_\_\_ Information Disclosure Statement.

\_\_\_\_\_ Attached Form 1449.

\_\_\_\_\_ Copies of each of the references listed on the attached Form PTO-1449 are enclosed herewith.

\_\_\_\_\_ Petition for Extension of Time for parent application, Together with the appropriate extension fee.

\_\_\_\_\_ Appended Materials as follows: \_\_\_\_\_

☒ Return Receipt Postcard specifically itemized.

\_\_\_\_\_ Other as follows:

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FEE CALCULATION	<u>BASIC FEES</u>
Provisional Application	\$150.00
Design Application	\$310.00
Utility Application	\$760.00

For	Number Filed	Number Extra	Rate	Fee
Basic Fee				\$ 760.00
Total Claims	18 - 20	0	x \$18.00	= \$ 0.00
Independent Claims	2 - 3	0	x \$78.00	= \$ 0.00
Multiple Dependent Claims			+ \$260.00	= \$ 0.00
TOTAL FILING FEES				\$ 760.00

X A Check is enclosed in the amount of \$760.00.

X The Commissioner is authorized to charge payment of the following fees and to refund any overpayment associated with this communication or during the pendency of this application to deposit account 14-0930. This sheet is provided in duplicate.

\_\_\_\_\_ The foregoing amount due.

X Any additional filing fees required, including fees for the presentation of extra claims under 37 C.F.R. 1.16.


X Any additional patent application processing fees under 37 C.F.R. 1.17 or 1.20(d).

\_\_\_\_\_ The issue fee set in 37 C.F.R. 1.18 at the mailing of the Notice of Allowance.

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SHOULD ANY DEFICIENCIES APPEAR which respect to this application, including deficiencies in payment of fees, missing parts of the application or otherwise, the United States Patent and Trademark Office is respectfully required to promptly notify the undersigned.

Date: September 1, 1999

  
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## CAST MEMBRANE STRUCTURES FOR SAMPLE PREPARATION

### BACKGROUND OF THE INVENTION

A number of analytical procedures have been developed in the biochemical art wherein it is required to remove solvent from peptide solutions in order to have a more concentrated peptide sample which can be analyzed effectively, or to remove low molecular weight ions or solutes. Many other analytical procedures, involving not only peptides but macromolecular species in general, also have been developed wherein it is necessary to concentrate and/or "desalt" a macromolecular component in a liquid sample, as there is commonly a need in biochemistry/medicinal chemistry for pure analytes devoid of salts, detergents and other contaminants. The presence of these substances can be deleterious, in that they often interfere with subsequent chemical analyses. Analogous situations exist in the environmental art and in chemical analysis.

U.S. Patent No. 4,755,301 discloses a centrifugal method and apparatus for concentrating macromolecules without filtering to dryness. A semipermeable ultrafiltration membrane separates a sample reservoir from a filtrate cup, and filtrate ducts below the membrane are offset sufficiently inward from the edge of the membrane so that when the apparatus is used in a fixed angle centrifuge rotor, filtration stops once the retentate meniscus reaches the centrifugal radial level of the outermost edge of the outermost filtrate duct.

Such ultrafiltration devices are commonly used for the "purification" and/or sample preparation of biomolecules and natural products. For such a process to be successful, a

membrane must be selected that retains the molecules of interest, yet passes the impurities. Although this scenario is relatively straightforward for analytes greater than about 10,000 molecular weight, it becomes increasingly problematic for substances less than about 5000 molecular weight. The reason is due to the fact that the required membrane porosity to retain the about 5000 molecular weight analyte is so low that the water permeability (flow rate) becomes poor and processing times too long. For example, a typical centrifugal "spin time" for a device using a membrane suitable for analytes having a molecular weight of 30,000 or more is about one hour, whereas as many as six hours may be required for analytes of about 1000 molecular weight. Furthermore, such long term exposure to high g-forces frequently results in device failure.

The sample quantities now common in the art are in the 0.01 to 10 microgram range. At such low loads, efficient sample handling is crucial to avoid loss. Conventional methods and devices for sample preparation are not practical for handling the "microseparation" of such small sample volumes. In addition, ultrafiltration can only effectively concentrate and desalt, and thus the application of adsorption technology at this scale could offer an entirely new approach to micro-mass sample preparation.

One conventional method for making sample preparation devices is to first insert a precut porous plug obtained from, for example, a fibrous glass or cellulose sheet, into the tip of a pipette, followed by the addition of loose particles and a second porous plug, as illustrated in Figure 1. The plugs serve to retain the particles in place in the pipette tip. However,

the plugs also entrap excess liquid thereby creating dead space or volume (i.e., space not occupied by media or polymer that can lead to poor sample recovery, contamination such as by sample carry-over, etc.). However, these procedures cannot be used with extremely small liquid delivery devices such as pipette tips, as there is no practical way to load either the plug or the particles to obtain a microadsorptive device that contains 10 milligrams or less of adsorbent to be used for the aforementioned extremely small sample loads.

Alternatively, a micro sample preparation device can be made by lodging media in a capillary pipette. However, the flow through such devices is typically slow.

Moreover, although from a mass adsorption standpoint, adsorptive powders offer the highest capacity, they are difficult or indeed impossible to handle in milligram quantities. Although polymer-based adsorptive membrane sheets are relatively easy to handle, their capacity is poor as a result of relatively low substructure surface area.

It is therefore an object of the present invention to provide a sample preparation device which can concentrate, purify and/or desalt molecules from sample solutions.

It is another object of the present invention to provide a sample preparation device which can concentrate, purify and/or desalt molecules from very small sample solutions.

It is another object of the present invention to provide a sample preparation device which can concentrate, purify and/or desalt molecules from sample solutions in a variety of form geometries.



It is a further object of the present invention to provide a sample preparation device which can concentrate, purify and/or desalt molecules from very small sample solutions in a variety of form geometries.

It is a still further object of the present invention to provide a sample preparation device that is simple and economic to manufacture.

It is yet a further object of the present invention to provide a method of casting particles in a housing in a variety of housing sizes or geometries.

It is a further object of the present invention to provide a castable membrane that assumes the shape of the housing in which it is cast, and can be retained in that housing without the use of porous plugs.

It is another object of the present invention to provide a castable membrane on a support or substrate.

#### SUMMARY OF THE INVENTION

The problems of the prior art have been overcome by the present invention, which provides a method for casting in-place composite (filled) and/or non-filled structures which are useful as sorptive or reactive media or for size-based separations. In one embodiment, the structures are monolithic and/or continuous. The invention is applicable to a variety of particular housing sizes and configurations, and provides a means of affixing media in a housing of a variety of volumes. The invention enables the inclusion of a substantial (relative to the increase in surface area of the precipitated structure) amount of media in the

polymer while still retaining a three dimensional polymeric structure.

In a first preferred embodiment, the composite structures comprise particles entrapped within a porous polymeric substrate, such as that shown in Figure 2B, and are cast in-place into a housing of a variety of sizes, such as a pipette tip as illustrated in Figure 2A, thereby providing an effective platform for micromass handling. With the appropriate selection of particle chemistry, virtually any separation or purification operation can be conducted, including selective bind/elute chromatography operations, on sample mass loads less than 1 microgram in volumes of a few microliters, as well as larger mass loads and volumes. The present invention also encompasses the composite structures as well as sample preparation devices containing the same.

In a second preferred embodiment, unfilled structures which may be self-retaining and/or self-supporting are cast *in situ* in a suitable housing and can be used for size-based separations wherein the cast structure acts as a semi-permeable barrier, or for adsorption. The present invention also encompasses these structures as well as housings containing these structures, such as that shown in Figure 3. The device in Figure 3 is a centrifugal device having a reservoir, a base, and a porous fabric sealed between the reservoir and base. The structures of the invention are cast-in-place on the porous fabric. The device is placed in a suitable vial during operation, and the flux of the device is driven by centrifugal force.

## BRIEF DESCRIPTION OF THE DRAWINGS

Figure 1 is a schematic diagram of an adsorptive pipette tip assembled with particles between two porous plugs;

Figure 2A is a schematic diagram of an adsorptive pipette tip prepared by a cast-in-place process in accordance with the present invention;

Figure 2B is a scanning electron micrograph of a particle loaded cast-in-place structure;

Figure 3 is an further embodiment of a housing into which a cast-in-place structure is added;

Figure 4 is a scanning electron micrograph of a particle loaded membrane prepared with spherical silica gel and polysulfone binder;

Figure 5A illustrates a multi-well array as a suitable housing for the cast-in-place structures of the present invention;

Figure 5B is a side view of an underdrain that can be used with the multi-well array of Figure 5A;

Figure 5C is a top view of a single well of the multi-well array of Figure 5A;

Figure 5D is a cross-sectional view of a well of the underdrain of Figure 5B;

Figure 6 is a reversed phase chromatogram of a 5 peptide mixture;

Figure 7 is a reversed phase chromatogram of a 5 peptide mixture which was bound and eluted from a pipette tip containing cast-in-place C18 silica;

Figure 8 is a reversed phase chromatogram of a 5 peptide

904000-ETH-3000  
mixture which was bound and eluted from a pipette tip containing cast-in-place styrene sulfonate coated silica;

Figure 9 is a reversed phase chromatogram of cytochrome c;

Figure 10 is a reversed phase chromatogram of cytochrome c after a 15 minute exposure to a pipette tip containing cast-in-place immobilized trypsin beads;

Figure 11 is an electrophoretic gel of a rabbit immunoglobulin-g sample that has been bound and eluted from a pipette tip containing cast-in-place immobilized Protein A beads;

Figure 12 is an electrophoretic gel of supercoiled plasmid DNA which has been bound and eluted from a 1000  $\mu$ l pipette tip containing cast-in-place silica;

Figure 13 is an electrophoretic gel of linear DNA fragments ranging from 100-2000 bp which have been bound and eluted from a 200  $\mu$ l pipette tip containing cast-in-place silica;

Figure 14 is an electrophoretic gel of PCR amplified DNA which has been bound and eluted from a 200  $\mu$ l pipette tip containing cast-in-place fumed silica;

Figure 15 is an electrophoretic gel of linear DNA fragments which have been bound and eluted from a 200  $\mu$ l pipette tip containing loose silica and a cast-in-place membrane barrier;

Figure 16A is a scanning electron micrograph of a longitudinal section of a pipette tip containing a cast-in-place particle loaded membrane prepared with spherical silica gel and polysulfone binder; Figure 16B is a scanning electron micrograph of a cross-section of a pipette tip containing a cast-in-place particle loaded membrane prepared with spherical silica gel and polysulfone binder; and

Figure 17 is a longitudinal section of a pipette tip containing a cast-in-place unfilled uncut membrane.

#### DETAILED DESCRIPTION OF THE INVENTION

The term "membrane" as used herein includes permeable and semi-permeable three dimensional structures with or without particles, having a porosity suitable for the desired application. The term "composite structure" as used herein includes filled membranes.

In the first preferred embodiment of the present invention, those skilled in the art will recognize that many different particles can be used in the composite structures, depending upon the desired objectives of the resulting device. In the case of adsorptive devices, the ideal device will have rapid adsorption kinetics, a capacity and selectivity commensurate with the application, and allows for elution of bound analyte with an appropriate desorption agent. Suitable adsorptive composite structures are polymer bound, particle laden adsorptive membrane structures, such as those comprised of chromatographic beads which have been adhered together with a binder. A suitable polymer bound particle laden adsorptive membrane is illustrated in Figure 4. This membrane is comprised of about 80% w/w silica and 20% w/w polysulfone binder, and is produced by Millipore Corporation. A similar membrane is shown in Figure 16A cast-in-place in a pipette tip 50. Functional composite structures comprising other micron-size (e.g., 1-30 microns) resin particles derivatized with other functional groups are also beneficial,

including styrene-divinyl-benzene-based media (unmodified or derivatized with e.g., sulphonic acids, quaternary amines, etc.); silica-based media (unmodified or derivatized with C<sub>2</sub>, C<sub>4</sub>, C<sub>6</sub>, C<sub>8</sub>, or C<sub>18</sub> or ion exchange functionalities), to accommodate a variety of applications for peptides, proteins, nucleic acids, and other organic compounds. Those skilled in the art will recognize that other matrices with alternative selectivities (e.g., hydrophobic interaction, affinity, etc.) can also be used, especially for classes of molecules other than peptides. The term "particles" as used herein is intended to encompass particles having regular (e.g., spherical) or irregular shapes, as well as shards, fibers and powders, including metal powders, plastic powders (e.g., powdered polystyrene), normal phase silica, fumed silica and activated carbon. For example, the addition of fumed silica into a polysulfone polymer results in increased active surface area and is suitable for various applications. Polysulfone sold under the name UDEL P3500 and P1700 by Amoco is particularly preferred in view of the extent of the adherence of the resulting composite structure to polyolefin housing, including polypropylene, polyethylene and mixtures thereof. Other suitable polymer binders include polyethersulfone, cellulose acetate, cellulose acetate butyrate, acrylonitrile PVC copolymer (sold commercially under the name "DYNEL"), polyvinylidene fluoride (PVDF, sold commercially under the name "KYNAR"), polystyrene and polystyrene/acrylonitrile copolymer, etc.. Adhesion to the housing can be enhanced or an analogous effect achieved with these composite structures by means known to those skilled in the art, including etching of the housing, such as with plasma

treatment or chemical oxidation; mechanical aids such as rims inside the housing; and inclusion of additives into the housing material that promote such adhesion. Adhesion allows uniform precipitation during casting.

Devices in accordance with the present invention may incorporate a plurality of composite structures having resin materials with different functional groups to fractionate analytes that vary by charge, size, affinity and/or hydrophobicity; alternately, a plurality of devices containing different individual functional membranes may be used in combination to achieve a similar result. Similarly, one or more membranes can be cast in a suitable housing and functionality can be added before or after casting.

In accordance with the present invention, the structures of the present invention can be formed by a polymer phase inversion process, air casting (evaporation) and thermal inversion. For those systems with minimal or no adhesion, the formed structures can be separately prepared and inserted into the appropriate housing and held in place by mechanical means. In the preferred method, the formed structures are cast *in situ* in the desired housing. This results in the ability to include large amounts of media in the polymer matrix while still maintaining a three-dimensional porous structure. The membrane substructure serves as a support network enmeshing the particles, thus eliminating the need for frits or plugs, thereby minimizing or even eliminating dead volume (the adsorptivity of the membrane may or may not participate in the adsorption process). However, porous frits plugs could be added if desired. Preferably the membranes

or composite structures formed have an aspect ratio (average diameter to average thickness) of less than about 20, more preferably less than about 10, especially less than 1. For example, for adsorptive pipette tips, aspect ratios of two or less, more preferably less than 1 are preferred, especially between about 0.005-0.5. An aspect ratio within this range provides for suitable residence times of the sample in the composite structure during operation.

In the polymer phase inversion process, the solvent for the polymer must be miscible with the quench or inversion phase. For example, N-methyl-pyrrolidone is a suitable solvent for polysulfones, polyethersulfones and polystyrene. In the latter case, polystyrene pellets can be dissolved in N-methyl-pyrrolidone and case-in-place. The resulting structure shows good adhesion to the walls of a polyolefin-based housing, and has adsorption characteristics similar to polysulfone. Dimethylsulfoxide (DMSO), dimethylform-amide, butyrolactone, and sulfalane are also suitable solvents. N,N-dimethylacetamide (DMAC) is a suitable solvent for PVDF. Water is the preferred precipitant. The polymer phase inversion process generally results in an expansion of the structure to about two to three times its casting solution volume in the housing.

In the air casting process, a volatile solvent for the polymer binder is used. For example, in the case of cellulose acetate, acetone is a suitable volatile solvent. Air casting generally results in a structure which is smaller than the casting solution volume. With this method, particles in the filled structures should be at least about  $30\mu$  to allow flow



through the interstitial spaces after shrinkage without having to apply higher driving force.

The upper limit of particle amounts is dictated by casting solution viscosity. Depending on particle type, up to 40% (w/w) of particles can be added to the polymer without resulting in a casting solution too viscous to draw into the housing. Higher particle loadings may be achieved using higher temperature. Suitable particle sizes include particles in the range of from about 100 nanometers to about 100 microns in average diameter with or without porosity.

Suitable housing materials are not particularly limited, and include plastics (such as polyethylene and polypropylene), glass and stainless steel. Polyolefins, and particularly polypropylene, are preferred housing materials in view of the chemical adhesion that is created with the composite structure when the composite containing polysulfone, and in particular UDEL P3500 and P1700 polysulfones available from Amoco, is cast-in-place therein. Figure 16B illustrates such adhesion with a polypropylene pipette tip housing having a cast-in-place membrane therein prepared with spherical silica gel and polysulfone.

Suitable housing configurations are also not particularly limited, and include pipette tips, wells, multi-well arrays, plastic and glass cavities, sample preparation devices such as the MICROCON<sup>o</sup> microconcentrator, commercially available from Millipore Corporation, etc.. The preferred housing configuration is substantially cylindrical, as the flow vectors during operation are substantially straight, similar to chromatography, thereby minimizing or avoiding dilutional washing that might

occur with non-cylindrical configurations. Although housings with volumes between about 0.1  $\mu$ l and about 5 mls. can be used for casting-in-place, volumes less than about 100 $\mu$ l are preferred, with volumes of from about 0.1-50 $\mu$ l, preferably from about 0.2-20 $\mu$ l, are especially preferred. Pipette tip geometries having volumes as small as about 5 microliters can be used. When chemical adhesion of the composite structure to the housing walls is desired but is insignificant or non-existent, mechanical means can be used to maintain the composite structure in the housing. such as crimping, press fitting, heat shrinking the housing or a portion thereof, plasma treating the housing or a portion thereof, or chemically treating, such as etching, the housing or a portion thereof to promote adhesion. An advantage of adhesion to the housing wall is the ability to "seal" the composite structure to the housing without mechanical means. Such sealing (by whatever method) prevents the sample from channeling or bypassing the composite during operation. Preferably the structures of the present invention have a final bed height of from about 0.05 to about 5mm. This allows for good washing, good density per unit volume, and results in a uniform precipitation during formation of the plug.

The structures of the present invention also can be cast-in-place in conventional multi-well arrays having suitable geometries. Alternatively, as shown in Figures 5A-5D, multi-well arrays 10 can be used as the housing, such as by casting the structures 11 of the present invention in place in the well 12. Alternatively, Figure 5B shows an underdrain subassembly 13 having a plurality of wells 12 (enlarged in Figure 5D) with cast-

in-place structures contained therein. The underdrain 13 can be adapted to be permanently or removably coupled to the reservoir array 10 by any suitable means, such as by snapping, so as to form removable "boot" assemblies containing the structures of the present invention. For convenience, each underdrain 13 can contain a polymer matrix having particles with different chemistry, so that the user chooses the appropriate underdrain 13 depending upon the application. Alternatively or in addition, the particle laden polymer matrix can differ from well to well. The reservoir housing 10 can be a plurality of open bores, or can include a membrane.

The composite structures and the micro sample preparation devices of the present invention containing the composite structures have a wide variety of applications, depending upon the particle selection. For example, applications include peptide and protein sample preparation prior to analysis, peptide removal from carbohydrate samples, amino acid clean-up prior to analysis, immobilized enzymes for micro-volume reactions, immobilized ligands for micro-affinity chromatography, isolation of supercoiled and cut plasmids, clean-up of PCR and DNA products, immobilized oligo dT for RNA isolation, dye terminator removal, sample preparation for elemental analysis, etc.. Those skilled in the art will be able to choose the appropriate particles, polymer binder, particle chemistry and form geometry depending upon the desired application. In some cases, a mixture of particles can be used in the same devices. Alternatively or in addition, a multi-well device could have different chemistries for each separate well.

In the embodiment where the structures of the present invention are not filled with particles, symmetrical or asymmetrical semi-permeable structures, or a combination of symmetrical and asymmetrical semi-permeable structures, can be formed. In this embodiment, the preferred method of formation is casting *in situ* in the appropriate housing to form a self-retaining, self-supporting structure suitable for separations based on size or adsorption (depending on polymer identity). Functionality can be added to such a membrane to perform adsorption separations without the use of particles. For example, cellulose acetate can be treated with base to form cellulose, followed by an oxidant to render it reactive.

In the *in situ* formation process (either with filled or unfilled structures), the preferred method of formation involves precipitation by means of solvent exchange, such as by introducing the casting solution into the housing by any suitable means, such as where pressure is the driving force, for example by capillary action or by using a vacuum source. In the embodiment in which the housing is a pipette tip, a preferred driving force is a hand-held pipettor. Once the desired volume in the housing is filled with casting solution, the casting solution in the housing is contacted with a liquid in which the polymer is insoluble, preferably water, so that the polymer precipitates in the housing. This can be accomplished by immersing the housing in the liquid, and/or drawing the liquid into the housing with a driving force such as by means of a vacuum. Through the exchange of water for the solvent, the structure precipitates. Those skilled in the art will appreciate

that the solvent used to prepare the casting solution and the non-solvent can contain a variety of additives.

At the first contact of the polymer with the precipitant, there is virtually instantaneous precipitation, thereby forming a semi-permeable barrier or "skin". Such a barrier is illustrated in Figure 17 as element 60 in a housing 62. This barrier slows the rate of further precipitation of the substructure. Once precipitation is complete, the initial semi-permeable barrier 60 can be removed, such as by cutting the housing at a point above the barrier at a point above the barrier or by abrading exposed polymer. The semi-permeable barrier 60 can be optionally left in place to carry out size-based separations with unfilled structures, as the barrier acts as a micro-filtration membrane.

The cast in-place structure assumes the shape of the housing and results in a self-retaining homogeneous structure akin to a chromatographic column, providing a large surface area suitable for bind/elute chromatography (e.g., when particles are included in the polymer matrix) or for other analytical or biochemical techniques. Suitable driving forces include centrifugation, gravity, pressure or vacuum.

Without limitation, the following examples illustrate the objects and advantages of the present invention.

Example 1:        Strong Cation Exchange (SCX) Silica in 20  $\mu$ l  
                 Pipette Tips

In a suitable small vessel, 5 grams of a 7 % (w/w) PVDF solution (Pennwalt Corp, KYNAR 761) was prepared in N,N-dimethylacetamide. To this, 1 gram of SCX, 200 Å, 15  $\mu$ m (Millipore, PN 85864) spherical silica was added and mixed

thoroughly with a spatula. The mixture was allowed to equilibrate for 2 hours at room temperature, then mixed again. A 20  $\mu$ l fluted polypropylene disposable pipette tip was affixed to a common P-20 Pipetman (Gilson, Ranin, etc.) and the volume adjustment was set to 20  $\mu$ l. The plunger was depressed to the bottom and the end of the pipette was placed into the casting solution. While carefully watching, the plunger was slowly raised to fill the tip with ca. 0.5 - 1  $\mu$ l of casting solution. Once the tip contained sufficient liquid, equal pressure was maintained, and the pipette tip was removed and dipped into a bath of deionized water @ 60° C for ca. 5 seconds. After this brief period, pressure was released on the plunger and water was drawn into the tip to precipitate the polymer. When the water level was ca. 0.5 cm above the polymer height, the tip was ejected into the bath and solvent exchange was allowed to occur for ca. 5 minutes. The tip was removed from the water bath and any precipitated polymer located on the exterior was abraded off. The tip was re-affixed to the pipettor and the liquid expelled. If the flow is poor, ca. 0.25 mm can be cut off the end with a sharp razor blade. To ensure that all solvent was removed, ca. 5 to 20  $\mu$ l of deionized water was drawn in and expelled several times.

Example 2: C18 Silica in Common 200  $\mu$ l Pipette Tips

In a suitable small vessel, 5 grams of a 6% (w/w) polysulfone solution (Amoco, P3500) was prepared in N-methyl-2-pyrrolidone. To this 2 grams of C18, 200 Å, 15  $\mu$ m spherical silica (Millipore, PN 85058 ) was added and mixed thoroughly with a spatula. The mixture was allowed to equilibrate for 2

hours at RT., then mixed again. A 200  $\mu$ l fluted polypropylene disposable pipette tip was affixed to a common P-200 Pipetman (Gilson, Ranin, etc.) and the volume adjustment was set to 200  $\mu$ l. The plunger was depressed to the bottom and the end of the pipette was placed into the casting solution. While carefully watching, the plunger was slowly raised to fill the tip with ca. 2 - 5  $\mu$ l of casting solution. Once the tip contained sufficient liquid, equal pressure was maintained, and the tip was removed and dipped into a bath of deionized water at room temperature for ca. 5 seconds. After this brief period, pressure on the plunger was released and water was drawn into the tip to precipitate the polymer. When the water level was ca. 0.5 - 1 cm above the polymer height, the tip was ejected into the bath and solvent exchange was allowed to occur for ca. 5 minutes. The tip was removed from the water bath and any precipitated polymer located on the exterior was twisted off. The tip was re-affixed to the pipetter and the liquid expelled. If the flow is poor, ca. 0.5 mm can be cut off the end with a sharp razor blade. To ensure that all solvent was removed, ca. 50 to 200  $\mu$ l of deionized water was drawn in and expelled several times.

Example 3:      60 Å, 10  $\mu$ m Normal Phase Silica in Wide Bore  
                  1000 $\mu$ l Pipette Tips

In a suitable small vessel, 6 grams of a 6% (w/w) cellulose acetate solution (Eastman Kodak, 398-60) was prepared in N-methyl-2-pyrrolidone. To this, 1 gram of 60 Å, 10  $\mu$ m granular silica gel (Davison, Grade 710) was added and mixed thoroughly with a spatula. The mixture was allowed to

equilibrate for 2 hours at room temperature, then mixed again. A wide bore 1000  $\mu$ l polypropylene pipette was affixed to a common P-1000 Pipetman (Gilson, Ranin, etc.) and the volume adjust was set to 1000  $\mu$ l. The plunger was depressed to the bottom and the end of the pipette was placed into the casting solution. While carefully watching, the plunger was slowly raised to fill the tip with ca. 10 - 25  $\mu$ l of casting solution. Once the tip contained sufficient liquid, equal pressure was maintained, and the tip was removed and dipped into a bath of deionized water for ca. 5 seconds. After this brief period, pressure on the plunger was released and water was drawn into the tip to precipitate the polymer. When the water level was ca. 1 cm above the polymer height, the tip was ejected into the bath and solvent exchange was allowed to take place for ca. 5 minutes. The tip was removed from the water bath and any precipitated polymer located on the exterior was abraded off. The tip was re-affixed to the pipettor and the liquid expelled. If the flow is poor, cut ca. 0.5 mm off the end with a sharp razor blade. To ensure that all solvent was removed, ca. 200 to 1000  $\mu$ l of deionized water was drawn in and expelled.

#### Example 4: Fumed Silica in Wide Bore 200 $\mu$ l Pipette Tips

In a suitable small vessel, 8 grams of a 7.5 % (w/w) polysulfone solution (Amoco, P3500) was prepared in N-methyl-2-pyrrolidone. To this, 0.5 grams of fumed silica (Degussa, Aerosil 200) were added and mixed thoroughly with a spatula. The mixture was allowed to equilibrate for 2 hours at room



temperature, then mixed again. A 200  $\mu$ l wide bore polypropylene pipette was affixed to a common P-200 Pipetman (Gilson, Ranin, etc.) and the volume adjust was set to 200  $\mu$ l. The plunger was depressed to the bottom and the end of the pipette was placed into the casting solution. While carefully watching, the plunger was slowly raised to fill the tip with ca. 10 - 25  $\mu$ l of casting solution. Once the tip contained sufficient liquid, equal pressure was maintained, and the tip was removed and dipped into a bath of deionized water for ca. 5 seconds. After this brief period, pressure on the plunger was released and water was drawn into the tip to precipitate the polymer. When the water level was ca. 1 cm above the polymer height, the tip was ejected into the bath and solvent exchange was allowed to take place for ca. 5 minutes. The tip was removed from the water bath and any precipitated polymer located on the exterior was abraded off. The tip was re-affixed to the pipettor and the liquid expelled. If the flow is poor, cut ca. 0.5 mm off the end with a sharp razor blade. To ensure that all solvent was removed, ca. 200 to 1000  $\mu$ l of deionized water was drawn in and expelled.

Example 5:      C18, 15  $\mu$ m silica particle loaded membrane cast in place

In a small vessel, 5 grams of a 6 % (w/w) polysulfone solution (Amoco, P3500) was prepared in N-methyl-2-pyrrolidone. To this, 2 grams of C18, 200 Å, 15  $\mu$ m silica (Millipore, PN 85864) was added and mixed thoroughly with a spatula. The mixture was allowed to equilibrate for 2 hours at room temperature, then mixed again. Using a pipette or eye

dropper, 25 - 50  $\mu$ l of casting solution was dispensed into a suitable fixture. Examples of such devices include (but are not limited to) an Millipore Microcon or the wells of a 96 well filter plate. When preparing devices by this method, each chamber must contain a permeable barrier which will retain the solution (e.g. polypropylene fabric, membrane, etc.). Once added, the unit was gently tapped to ensure that the solution covered the entire barrier surface. The device was immersed in water for ca. 2 hours, and was gently stirred every 15 mins to promote solvent exchange. After this period, the units were removed and placed in either a centrifuge or vacuum manifold, as appropriate. The cast in place structure was flushed with 500 to 1000  $\mu$ l of deionized water to ensure solvent removal.

Example 6.

Cast Porous End Plug in Wide Bore 1000  $\mu$ l  
Pipette Tips containing Loose 30  $\mu$ l Silica

In a suitable small vessel, 5 grams of a 7.5 % (w/w) polysulfone solution (Amoco, P3500) was prepared in N-methyl-2-pyrrolidone. The mixture was allowed to equilibrate for 2 hours at room temperature, then mixed again. A 1000  $\mu$ l wide bore polypropylene pipette was affixed to a common P-1000 Pipetman (Gilson, Ranin, etc.) and the volume adjust was set to 1000  $\mu$ l. The plunger was depressed to the bottom and the end of the pipette was placed into the casting solution. While carefully watching, the plunger was slowly raised to fill the tip with ca. 2 - 10  $\mu$ l of casting solution. Once the tip contained sufficient liquid, equal pressure was maintained, the tip was removed and dipped into a bath of

deionized water for ca. 5 seconds. After this brief period, pressure on the plunger was released and water drawn into the tip to precipitate the polymer. When the water level was ca. 0.5 cm above the polymer height, the tip was ejected into the bath and solvent exchange allowed to take place for ca. 5 minutes. The tip was removed from the water bath and any precipitated polymer located on the exterior was abraded off. The tip was re-affixed to the pipettor and the liquid expelled. If the flow is poor, cut ca. 0.5 mm off the end with a sharp razor blade. To ensure that all solvent was removed, ca. 100 to 500  $\mu$ l of deionized water was drawn in and expelled. The pipette was detached and any excess water in the upper chamber was removed with a cotton swab. 5 - 10 mg of (250 Å) 30  $\mu$ m silica gel was weighed out and carefully added to the back end of the pipette. The pipette was tapped so that the silica rested on top of the cast-in-place barrier. If necessary, affix a suitable porous plug (cotton or polypropylene) in the upper chamber to prevent particle loss.

#### Example 7: Cast Semi-Permeable Membrane Plug for Filtration

In a suitable vessel, 5 grams of 7.5% (w/w) polysulfone solution (Amoco, P3500) in N-methyl-2-pyrrolidone was prepared. The mixture is allowed to equilibrate for 2 hours at room temperature, and is then mixed again. A 1000  $\mu$ l wide bore polypropylene pipette is affixed to a common P-1000 Pipetman pipettor (Gilson, Ranin, etc.) and the volume adjust is set to 1000  $\mu$ l. The plunger is depressed to the bottom and the end of the pipette is placed into the casting solution.

While carefully watching, the plunger was slowly raised to fill the tip with ca. 2-10  $\mu$ l of casting solution. Once the tip contained sufficient liquid, equal pressure was maintained, and the tip was removed, excess polymer solution was wiped off, and the tip was dipped into a bath of deionized water for about 5 seconds. After this brief period, pressure was released on the plunger and water was drawn into the tip to precipitate the polymer. When the water level was about 0.5 cm above the polymer height, the tip was ejected into the bath and solvent exchange was allowed to take place for about 5 minutes. The tip was re-affixed to the pipettor, the liquid expelled, and washed with 100-200  $\mu$ l of deionized water. When cast in this manner, the precipitated polymer had a semi-permeable skin at the orifice, which can be used as a filtration medium.

Example 8: Porous 30  $\mu$ l Silica Plugs Prepared by Evaporation

In a suitable vessel, 5 grams of a 10% (w/w) cellulose acetate solution (Eastman Kodak, 398-60) in acetone was prepared. To this, 1 gram of methanol, 0.5 grams of deionized water and 1 gram of 250 Å, 30  $\mu$ m silica was added. The mixture was allowed to equilibrate for 2 hours at room temperature, and was then mixed again. A 1000  $\mu$ l wide bore polypropylene pipette was affixed to a common P-1000 Pipetman pipettor (Gilson) and the volume adjust was set to 1000  $\mu$ l. The plunger was depressed to the bottom and the end of the pipette was placed into the casting solution. The plunger was then slowly raised to fill the tip with about 5-10  $\mu$ l of

casting solution. Once the tip contained sufficient liquid, equal pressure was maintained, and the tip was removed, excess fluid was wiped off, and the tip was placed in a rack to allow solvent to evaporate for about 16 hours. After this period, the tip was washed with about 10  $\mu$ l of distilled water.

Example 9:                    30  $\mu$ l Silica End Plugs in Porous Polyethylene Prepared by Thermal Phase Inversion

In a suitable vessel, 5 grams of beaded polyethylene and 100 grams of mineral oil are added. The mixture is heated to 250°C on a hot plate with agitation. When the plastic liquifies, 4 grams of 250Å, 30  $\mu$ m silica is added and mixed thoroughly. Using a 1 ml graduated glass pipette with filler bulb, 50-100  $\mu$ l of the melt is drawn in. Once the tip contains sufficient liquid, equal pressure is maintained, and the tip is removed, excess plastic is wiped off, the tip is allowed to cool to room temperature. The pipette is transferred to a methylene chloride bath for 1 hour to extract the mineral oil. It is then removed, and the methylene chloride is expelled and allowed to air dry.

Example 10.                    C18 Silica 200  $\mu$ l Pipette Tips for Peptide Sample Preparation

Approximately 2.5  $\mu$ g of each peptide from a mixture consisting of GlyTyr (1), ValTyrVal (2), methionine enkephalin (3), leucine enkaphalin (4) and angiotensin II (5) (in 100  $\mu$ l 0.1 % TFA) was adsorbed to a P200 pipette tip containing ca. 5  $\mu$ l of cast C18, 200Å, 15  $\mu$ m spherical silica. The solution was drawn up and expelled 4 times. The tip was then washed with

200  $\mu$ l of 0.1% TFA. Bound peptides were eluted with 80% Acetonitrile in 0.1% TFA/water. The eluted peptides were diluted with 4 parts of 0.1 % TFA and analyzed by reverse phase HPLC (linear acetonitrile gradient 5 - 30 % over 20 min). The resulting chromatogram was then compared to that of the original mixture. (See Figures 6 and 7). As expected, the GlyTyr, ValTyrVal, which are small and relatively hydrophilic, did not bind to the  $C_{18}$ . The recoveries of the remaining 3 (adsorbed) peptides subsequent to elution ranged from 70-85 %.

Example 11. Strong Cation Exchange Silica 200  $\mu$ l Pipette Tips

Approximately 2.5  $\mu$ g of each solute from a mixture consisting of a five peptides (see Example 10) (in 100  $\mu$ l in 10 % glacial acetic acid) were adsorbed to a P200 pipette tip containing ca. 5  $\mu$ l of cast, styrene sulfonate coated, 300 Å, 15  $\mu$ m spherical silica. Adsorption was performed during 4 complete uptake-withdraw cycles followed by a 100  $\mu$ l flush with 20 % methanol/10 mM HCl. Bound sample was eluted with two 25  $\mu$ l volumes of 1.4 N ammonium hydroxide/50 % methanol. The eluted sample was analyzed by reversed phase HPLC and the resulting chromatogram was compared to that of the original mixture. (See Figures 6 and 8). The strong cation exchange tip bound all sample components, except GlyTyr. Such performance is consistent with the selectivity of sulfonic acid ion-exchange resins.

Example 12. Immobilized Enzyme in 200  $\mu$ l Pipette Tips

Trypsin was covalently coupled to an aldehyde activated

300 Å, 15 µm spherical silica and cast (20 µl) into P200 tips for protein digestion *in situ*. Trypsin activity within the tip was assessed by monitoring the digestion of cytochrome via HPLC. A sample of cytochrome c (10 µg in 100 µl of 100 mM Tris, 1 mM CaCl<sub>2</sub>, pH 8 @ 37C) was taken up into the tip for 15 minutes. The reaction was mixed 4X with a expel/draw cycle into an Eppendorf tube. The digest was analyzed by HPLC using a linear gradient of acetonitrile from 5 - 45% over 30 minutes (see Figure 10). The resulting chromatogram showed that greater than 90% of cytochrome c was digested after 15 minutes (see Figure 9 for undigested cytochrome c).

Example 13: Immobilized Protein A in 200 µl Pipette Tips

Recombinant protein A was coupled to precast P200 tips containing aldehyde-activated 300 Å, 15 µm spherical silica for the isolation of rabbit immunoglobulin (IgG). A 100 µl sample of 1 mg/ml IgG and BSA in RIP buffer (150 mM NaCl, 1% NP-40, 0.5% DOC, 0.1% SDS, 50 mM Tris, pH 8.0) was cycled six times through a tip containing 40 µl of cast volume containing protein A immobilized beads. The tip was then washed with 5 volumes of RIP buffer prior to the elution. Desorption of bound IgG was performed with (two 25 µl volumes) of 6M urea. The desorbed sample was diluted with 50 µl of 2X SDS Laemmli sample buffer and boiled for 3 min prior to electrophoretic analysis. This protocol was also performed on a blank tip containing just polysulfone without beads which served as a background control. Electrophoresis was performed in a 10-16 % acrylamide gel shown (see Figure 11). Samples are as

follows: Lane 9: (MW marker); lanes 1-4: increasing amounts of protein A tip eluted sample; and lanes 5-8: increasing amounts of eluted IgG/BSA from the blank polysulfone tip. These results indicate selective binding of IgG to the Protein A tip with minimal nonspecific adsorption. Furthermore, the blank tip (lanes 5-8), in the presence of detergents (RIP buffer), did not exhibit adsorption of either IgG or BSA.

Example 14: 60 Å, 10 µm 1000 µl Pipette Tips for Supercoiled DNA

*Escherichia coli* strain JM109 containing plasmid pUC19 was grown in 3 - 5 ml of Luria broth containing 100 µg/ml ampicillin at 37°C for 12-16 hours. 1.5 ml of the overnight culture was pelleted in a microfuge tube spun at maximum g-force for 30 sec at room temperature. Residual growth medium was removed while leaving the bacterial pellet intact. Plasmid DNA was then isolated using a modification of the alkaline lysis procedure of Birnboim and Doly (Birnboim, H. C. and Doly, J. (1979). *Nucleic Acids Res* 7., 1513). Briefly, the bacterial pellet was resuspended by vortexing in 50 µl of 50 mM glucose, 25 mM Tris-HCl (pH 8.0), 10 mM EDTA, and 10 µg/ml RNase A. Next 100 µl of 0.2 N NaOH, 1 % sodium dodecyl sulfate was added. The resulting suspension was incubated at room temperature for 2 min. Following the addition of 100 µl of 3 M sodium acetate solution (pH 4.8), the suspension was mixed by vortexing then spun in a microfuge at maximum g-force for 2 min. The cleared lysate was transferred to a fresh microfuge tube to which 7 M guanidine hydrochloride (GuHCl) in 200 mM 2-(N-morpholino)ethane sulfonic acid (MES) at pH 5.6



was added to a final concentration and volume of 4.4 M and 700  $\mu$ l, respectively. The resulting solution was drawn into a 1000  $\mu$ l polypropylene pipette tip with ca. 60  $\mu$ l of cast membrane containing ca. 60 Å, 10  $\mu$ m silica gel using a P-1000 pipettor. The solution was pipetted in-and-out for 2 - 2.5 minutes to allow extensive interaction between the DNA solution and the silica membrane matrix. The tip was then flushed once with 400  $\mu$ l of 80 % reagent grade alcohol. Residual alcohol is removed by repeated expulsion onto a paper towel. Plasmid DNA was eluted from the tip in 100  $\mu$ l of 10 mM Tris-HCl (pH 8.0), 1 mM EDTA (TE) by in-and-out pipetting 3x. Eluate fractions were adjusted to a final volume of 100  $\mu$ l with TE. Six tips were evaluated. To quantitate plasmid DNA recovery, 20 % of the eluate, as well as 20% of the unbound filtrates, were analyzed by agarose gel electrophoresis (See Figure 12). Included on the gel were samples of pUC19 plasmid DNA of known concentrations. (Lanes 1-4) Results of these experiments indicate that on average 2.5 mg of supercoiled plasmid was recovered (Lanes 5,7,9,11).

Example 15:      60 Å, 10  $\mu$ m Silica in Wide Bore 200  $\mu$ l Pipette  
                     Tips for Linear DNA

The ability of 200  $\mu$ l polypropylene wide bore pipette tips containing ca. 20  $\mu$ l of cast 60 Å, 10  $\mu$ m silica-laden membrane to bind linearized DNA fragments (pBR322 digested with either *Bst*NI or *Msp*I, to generate DNA fragment ladders) or plasmid pBR322 DNA restricted with *Pst*I and *Bam*HI (generates large linear restriction fragments) was assessed. Five  $\mu$ g of linearized plasmid DNA was combined with GuHCl, pH

5.6 in MES to a final concentration of 0.5 M and volume of 150  $\mu$ l. Prior to use, P-200 tips containing the silica membrane were pre-equilibrated in (2X) 200  $\mu$ l of 0.5 M GuHCl, pH 5.6 in MES. The DNA/GuHCl solution was drawn into a pipette tip and cycled in-and-out for 1.5-2.0 min to allow extensive interaction between the DNA binding mixture and the silica-laden membrane matrix. The tips were then washed with 125  $\mu$ l of 80% reagent grade alcohol to remove salts and other contaminants. Bound DNA was eluted from the tip matrix in 100  $\mu$ l TE, by in-and-out pipetting 3X. To measure DNA recovery, eluates and filtrates were analyzed by agarose gel electrophoresis (see Figure 13). In order to quantitate the amount of DNA recovered, samples representing 100%, 75%, 50%, and 25% of the starting material were run in lanes 1-4. Lanes 5, 7, 9, & 11 are the eluants. Estimate of band intensities indicate recoveries in excess of 95%.

Example 16. Fumed Silica in Wide Bore 200  $\mu$ l Pipette Tips for PCR Amplified DNA

The ability of 200  $\mu$ l wide bore polypropylene pipette tips containing ca. 20  $\mu$ l of fumed silica immobilized in a polysulfone matrix was assessed for the purification of PCR amplified DNA (500 bp). Prior to use, tips were flushed 2X with 100  $\mu$ l of TE buffer and then equilibrated with 500  $\mu$ l of 3 M NaI in 200 mM MES buffer (pH 6.4). 50  $\mu$ l samples from the pooled PCR stock (ca. 3  $\mu$ g of DNA) were then combined with 7 M NaI to a final NaI concentration of 3.0 M. The total volume following addition of the NaI solution was 150  $\mu$ l. The sample was drawn in and expelled from the P-200 tips containing the

cast fumed silica-laden membrane for 2-3 minutes allowing for extensive contact with the matrix. Each tip was then washed with 125  $\mu$ l of 80 % reagent grade alcohol to remove salts and other contaminants. Residual alcohol was removed by expelling the tip contents onto a paper towel. Bound PCR product was eluted in 50  $\mu$ l TE (pH 8.0). To estimate DNA recovery, eluates and filtrates were analyzed by agarose gel electrophoresis (see Figure 14). Loads representing 100%, 75%, 50%, and 25% of the starting material were run in lanes 1-4 as controls. Note the presence of the lower band which indicates a slight primer-dimer contamination. The use of immobilized fumed silica along with NaI appears to give an amplified DNA recovery in excess of 90 %. In addition, there appears to be a reduction in the primer-dimer contaminant. (See Lanes 5,7,9,11).

Example 17. Cast Porous End Plug with Loose 30 Micron Silica in a 200  $\mu$ l Pipette Tip for DNA Isolation

200  $\mu$ l pipette tips containing ca 5 - 10  $\mu$ l of cast (7.5 %) polysulfone as a porous end plug and 2 - 4 mg of loose 250 Å, 30  $\mu$ m silica was assayed for the ability to bind linear and supercoiled plasmid DNA. Regarding linear DNA, approximately 5  $\mu$ g of pBR322 was first digested with MspI in 45  $\mu$ l TE (10 mM Tris-HCl, 1 mM EDTA), pH 8.0, and then combined with 100  $\mu$ l of 7 M guanidine hydrochloride (GuHCl) in 200 mM MES buffer at pH 5.6. The final concentration of GuHCl in the solution was 4.7 M. The resulting solution was drawn (once) into a 200  $\mu$ l pipette tip and allowed to extensively contact the silica by

inverting the pipetman with the affixed tip for approximately 2 min. The DNA adsorbed to the tips was then washed and eluted as described in Example 15. Loads representing 100%, 75%, 50% and 25% of the starting material were run in Lanes 1-4 as controls. Results from experiments using this format indicate that DNA recoveries of better than 75% can be achieved (see Figure 15, Lanes 5 and 7).

What is claimed is:

1. A housing having an interior wall, said housing containing a structure comprising a plurality of sorptive particles bound to a polymer adhered to said interior wall.
2. The housing of claim 1, wherein said housing has a first open end and a second open end spaced from said first open end, and wherein said structure is contiguous with said second open end.
3. The housing of claim 1, wherein said housing defines a volume, and wherein said structure occupies about 0.1 microliters to about 10 milliliters of said volume.
4. The housing of claim 1, wherein said housing is a pipette tip having a first end and a second end spaced from said first end with said structure in between.
5. The housing of claim 4, wherein said first end has an internal diameter larger than the internal diameter of said second end.
6. The housing of claim 1, wherein said polymer is selected from the group consisting of polysulfone, polyethersulfone, polytetrafluoroethylene, cellulose acetate, polystyrene, polystyrene/acrylonitrile copolymer and PVDF.
7. The housing of claim 6, wherein said polymer is polysulfone, and wherein said housing comprises a polyolefin.
8. The housing of claim 7, wherein said polyolefin is polypropylene.
9. The housing of claim 1, wherein said particles are selected from the group consisting of polystyrenedivinylbenzene beads, functionalized polystyrenedivinylbenzene beads, silica, fumed silica, derivitized silica and activated carbon.
10. The housing of claim 1, wherein said particles are derivitized silica.
11. A housing having an interior wall, said housing containing a structure comprising an adsorptive polymer adhered to said interior wall.
12. The housing of claim 11, wherein said housing has a first open end and a second open

end spaced from said first open end, and wherein said structure is contiguous with said second open end.

13. The housing of claim 11, wherein said housing defines a volume, and wherein said structure occupies about 0.1 microliters to about 10 milliliters of said volume.

14. The housing of claim 11, wherein said housing is a pipette tip having a first end and a second end spaced from said first end with said structure in between.

15. The housing of claim 14, wherein said first end has an internal diameter larger than the internal diameter of said second end.

16. The housing of claim 11, wherein said polymer is selected from the group consisting of polysulfone, polyethersulfone, polytetrafluoroethylene, cellulose acetate, polystyrene, polystyrene/acrylonitrile copolymer and PVDF.

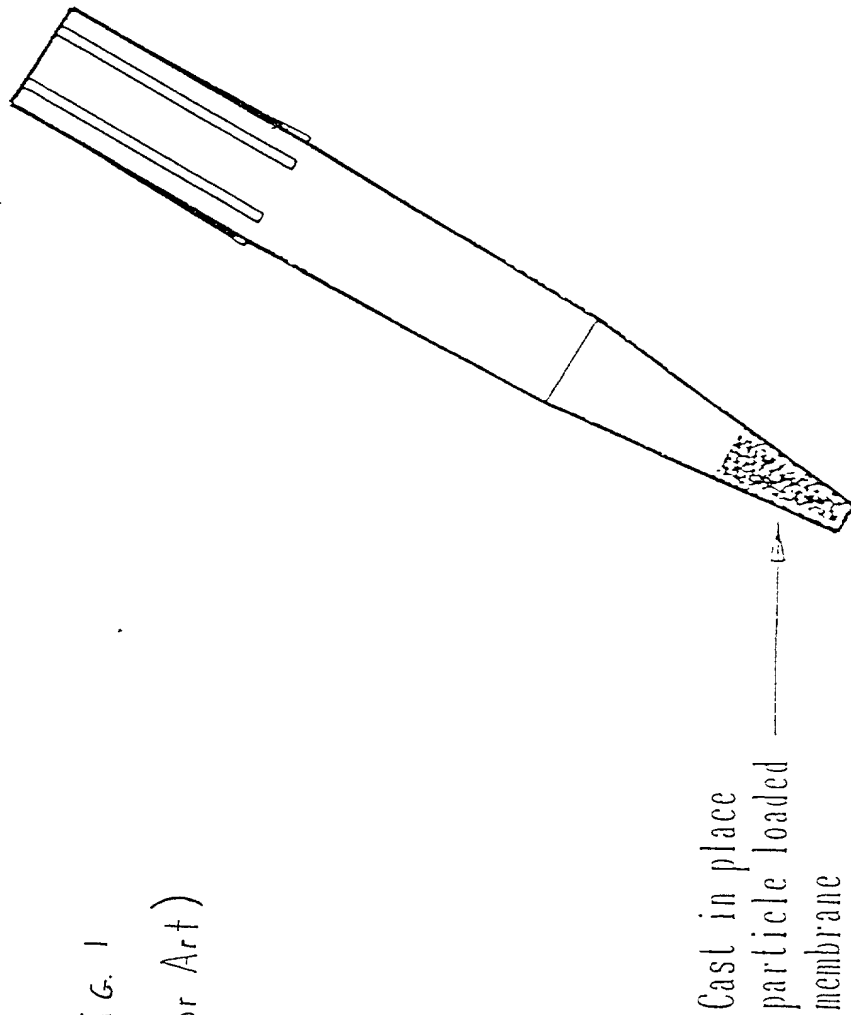
17. The housing of claim 16, wherein said polymer is polysulfone, and wherein said housing comprises a polyolefin.

18. The housing of claim 17, wherein said polyolefin is polypropylene.

# ABSTRACT

A method for casting-in-place composite and/or non-filled structures which are useful as sorptive or reactive media or for size-based separations. Any particular housing size or configuration can be used, and the inclusion of a large amount of adsorptive particles in polymer is achieved while still maintaining the membrane three dimensional structure. In a first preferred embodiment, the composite structures comprise particles entrapped within a porous polymeric substrate, and are cast in-place into a housing such as a pipette tip, thereby providing an effective platform for micromass handling. With the appropriate selection of particle chemistry, virtually any separation or purification operation can be conducted, including selective bind/elute chromatography operations, on sample mass loads less than 1 microgram in volumes of a few microliters, as well as larger mass loads and volumes. The present invention also encompasses the composite structures as well as sample preparation devices containing the same. In a second preferred embodiment, self-retaining, self-supporting structures are cast *in situ* in a suitable housing and can be used for size-based separations wherein the cast structure acts as a semi-permeable barrier. The present invention also encompasses these structures as well as housings containing these structures.

Fig. 1  
(Prior Art)





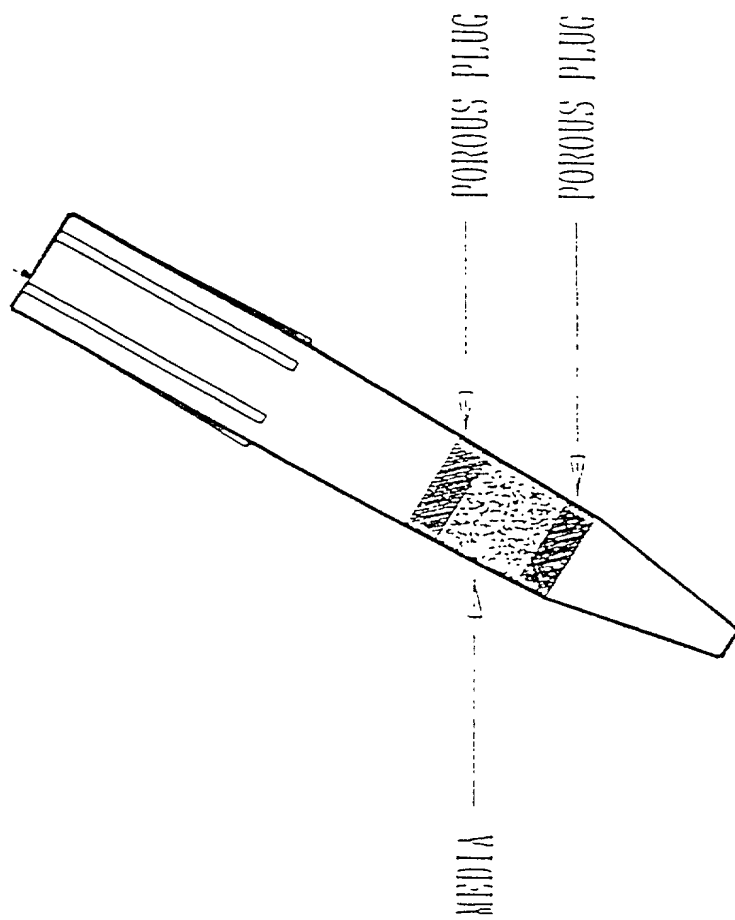


FIG. 2A

Fig. 2B

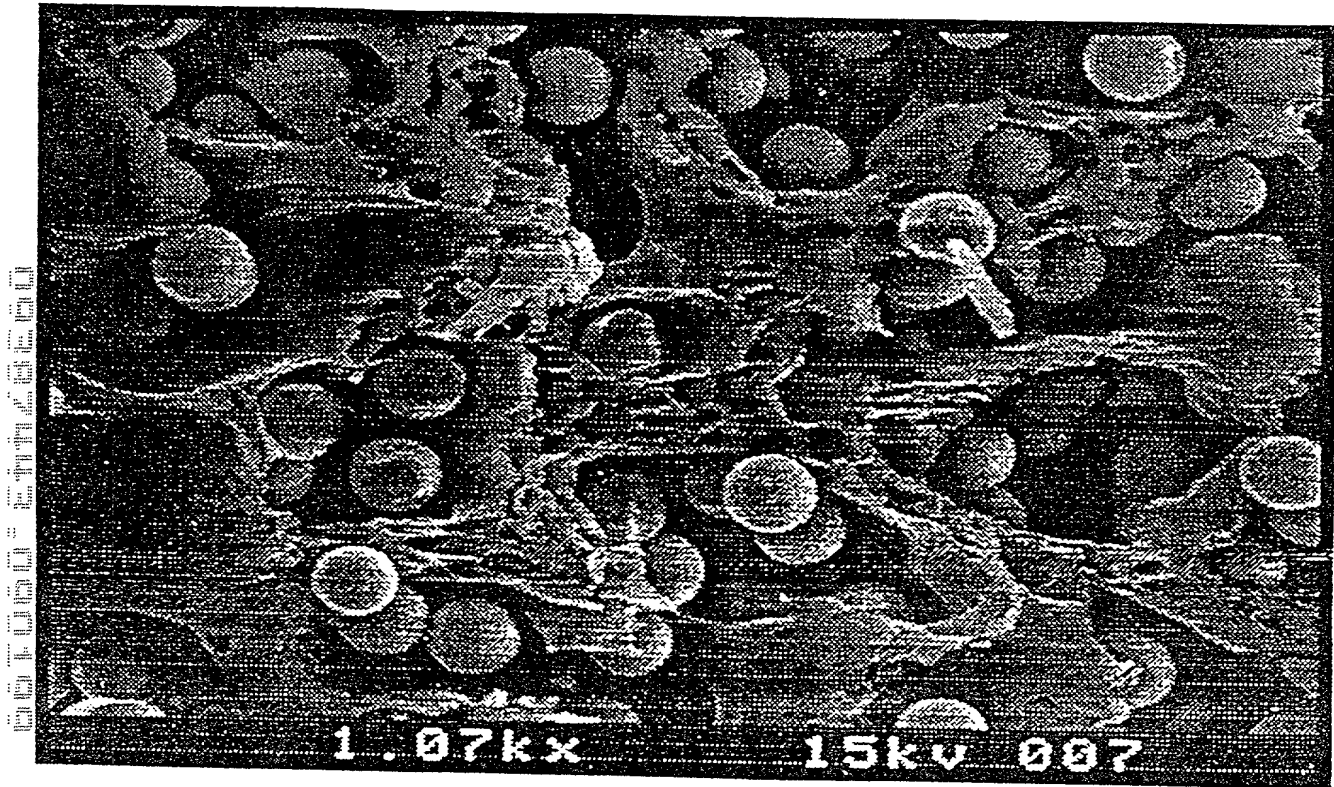


FIG. 3

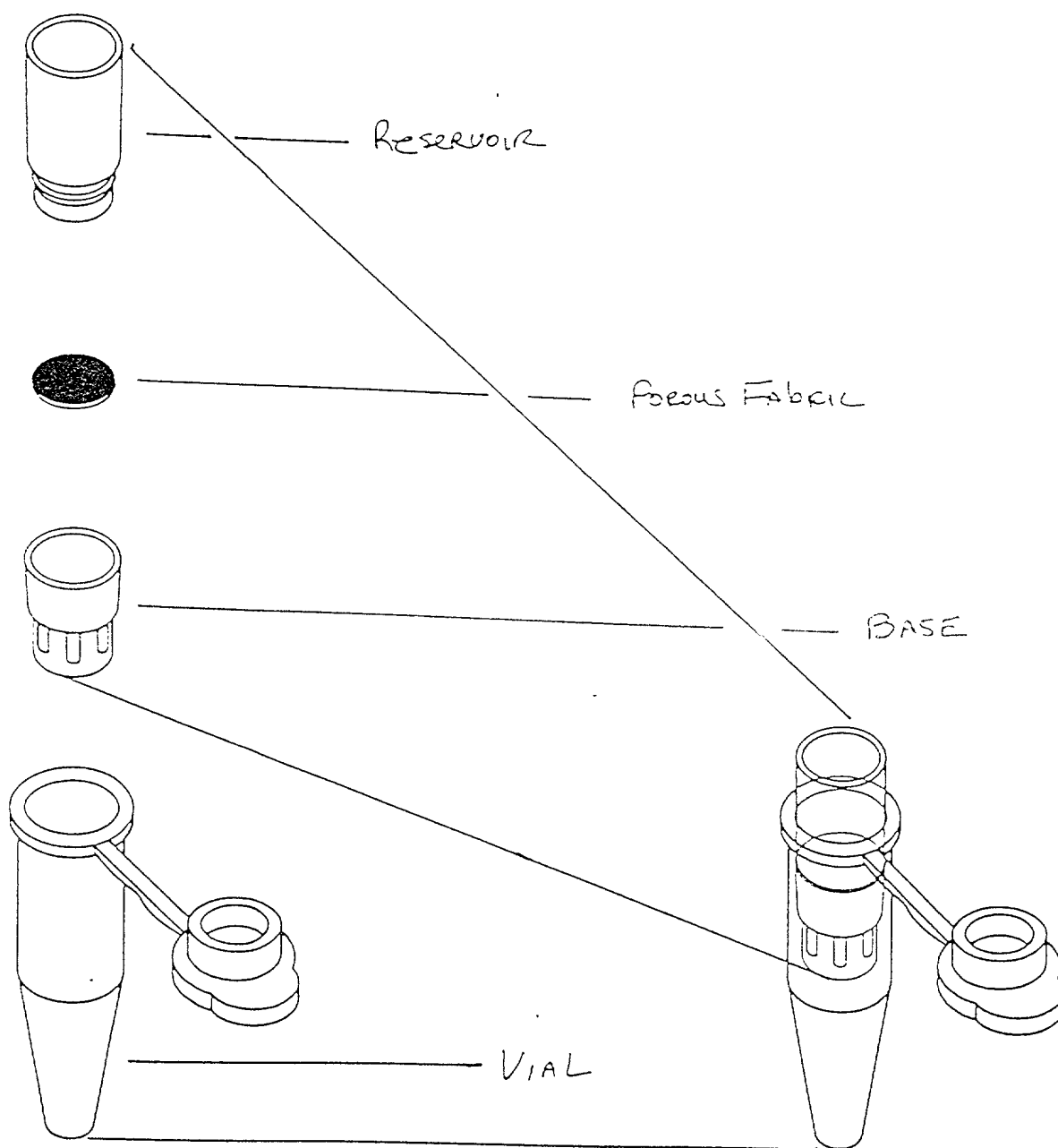


Fig. 4

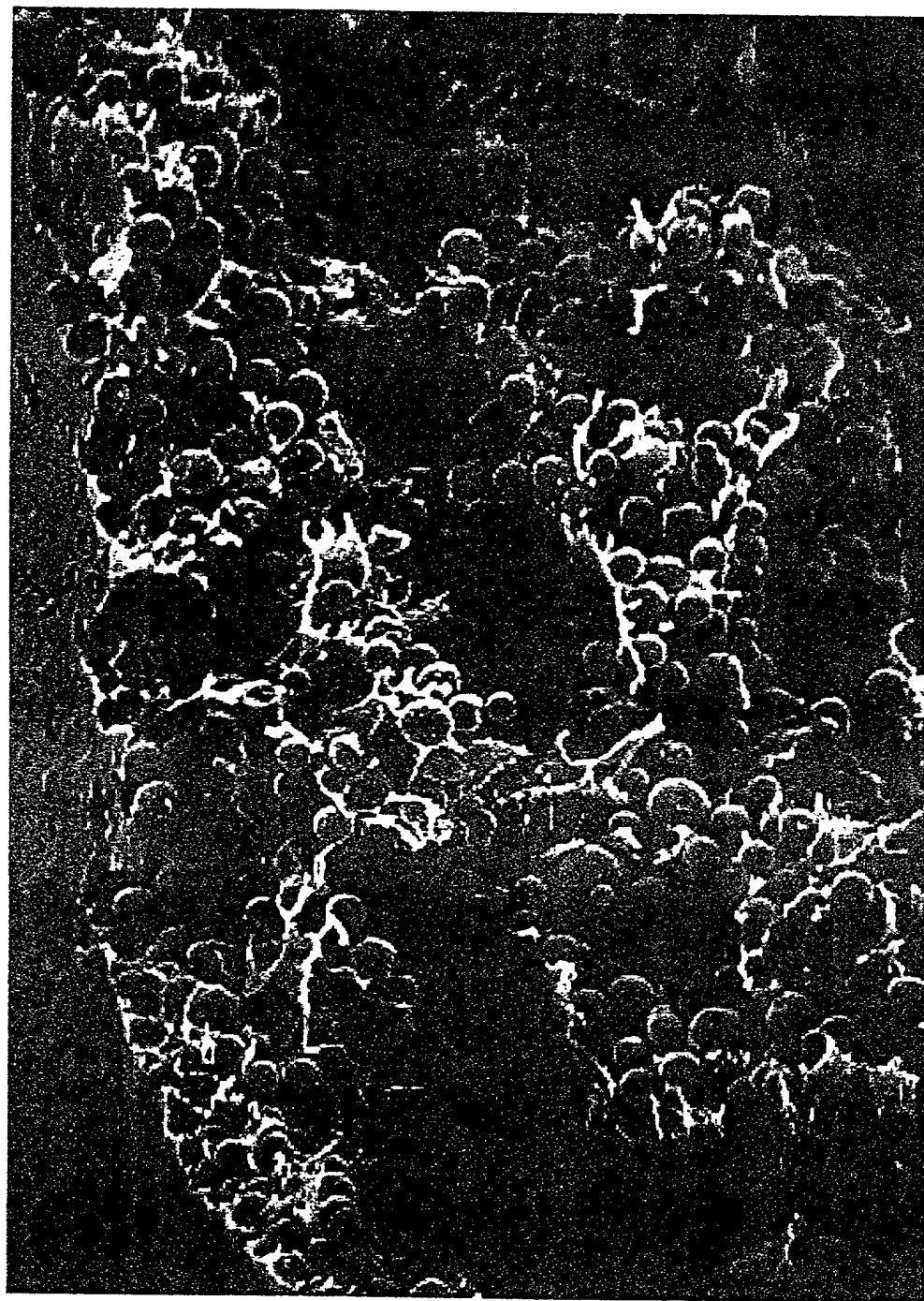


Fig. 5A

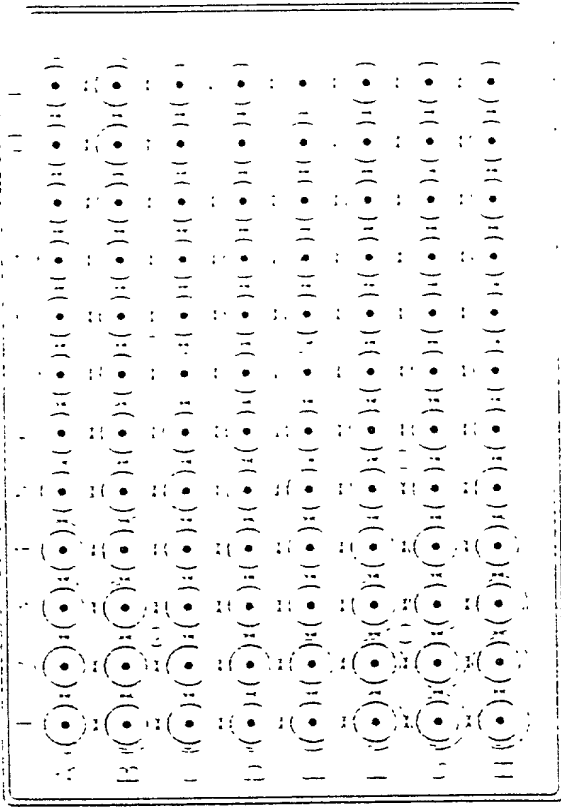


Fig. 5C

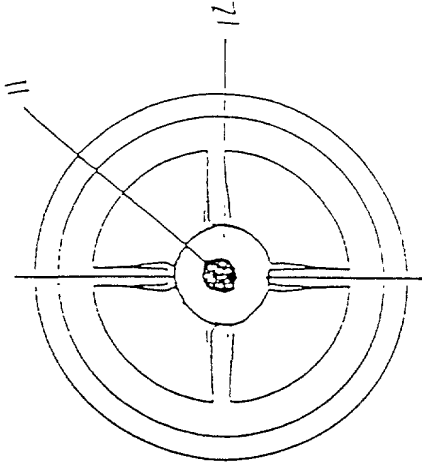


Fig. 5B

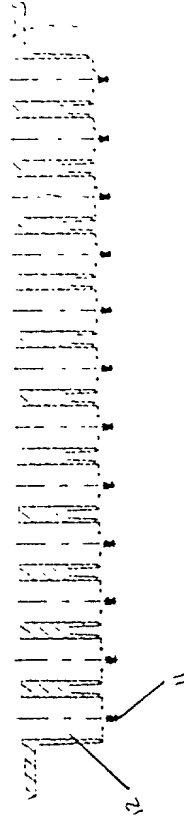


Fig. 5D

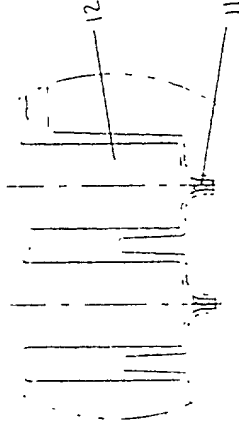


Fig. 6

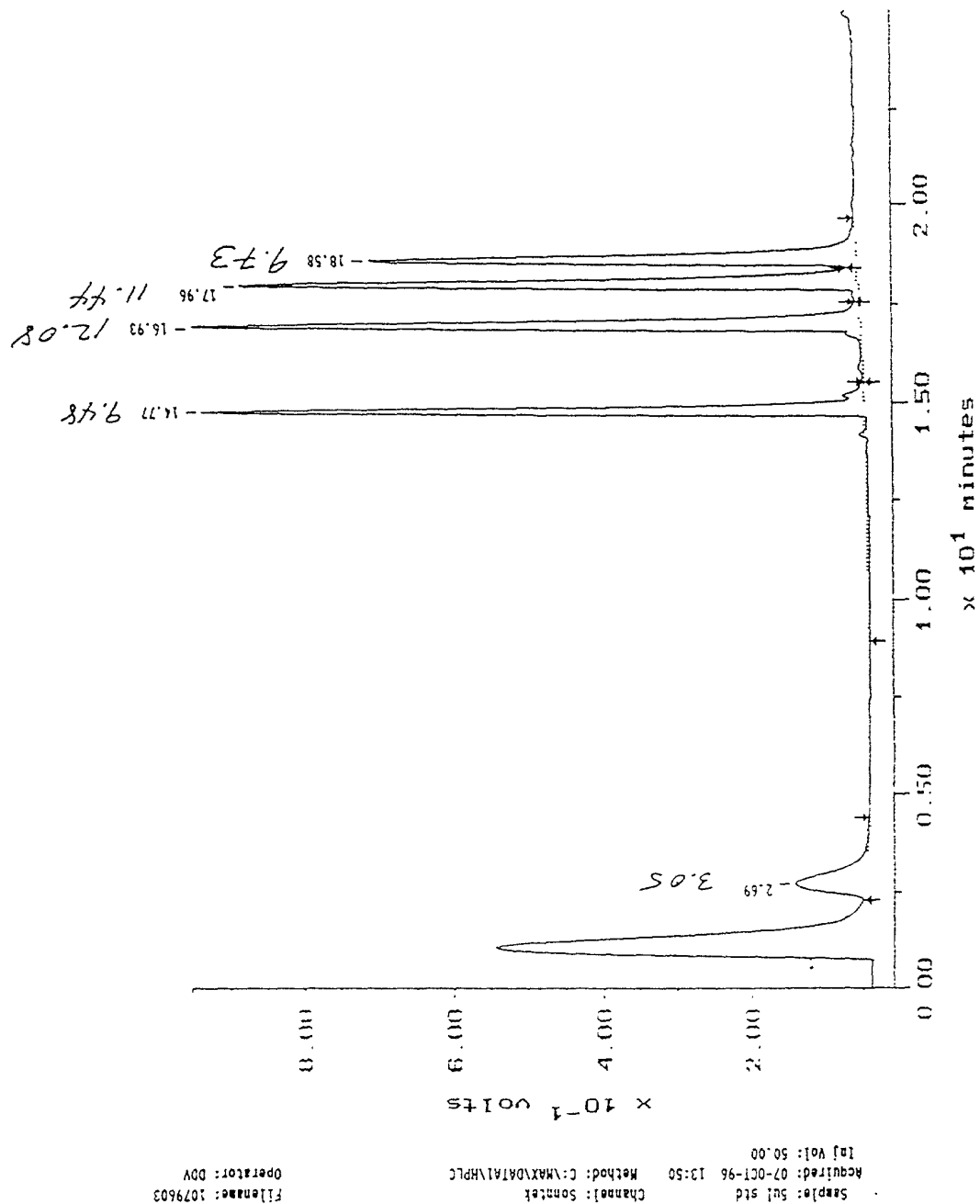


FIG. 7

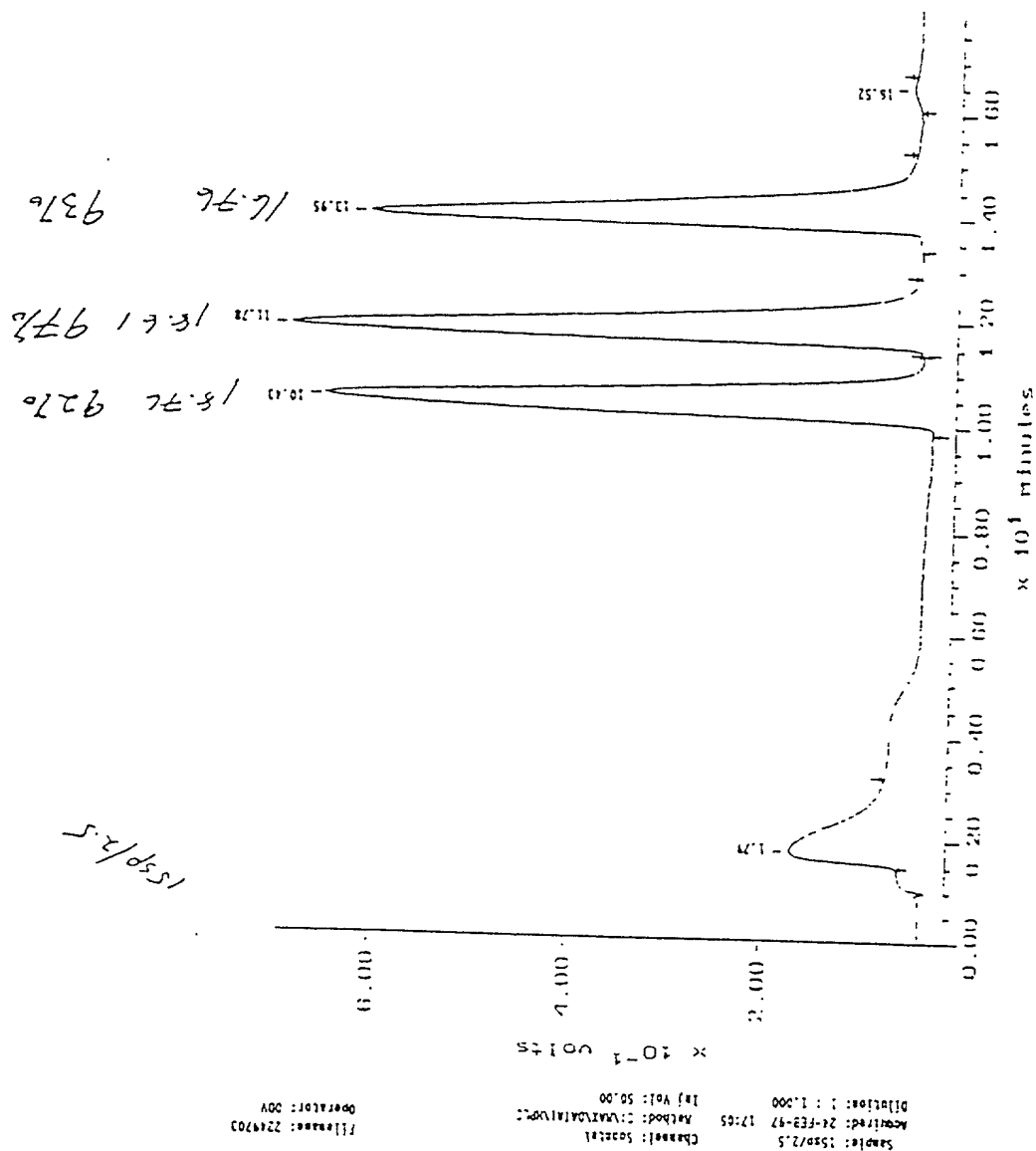


Fig. 8

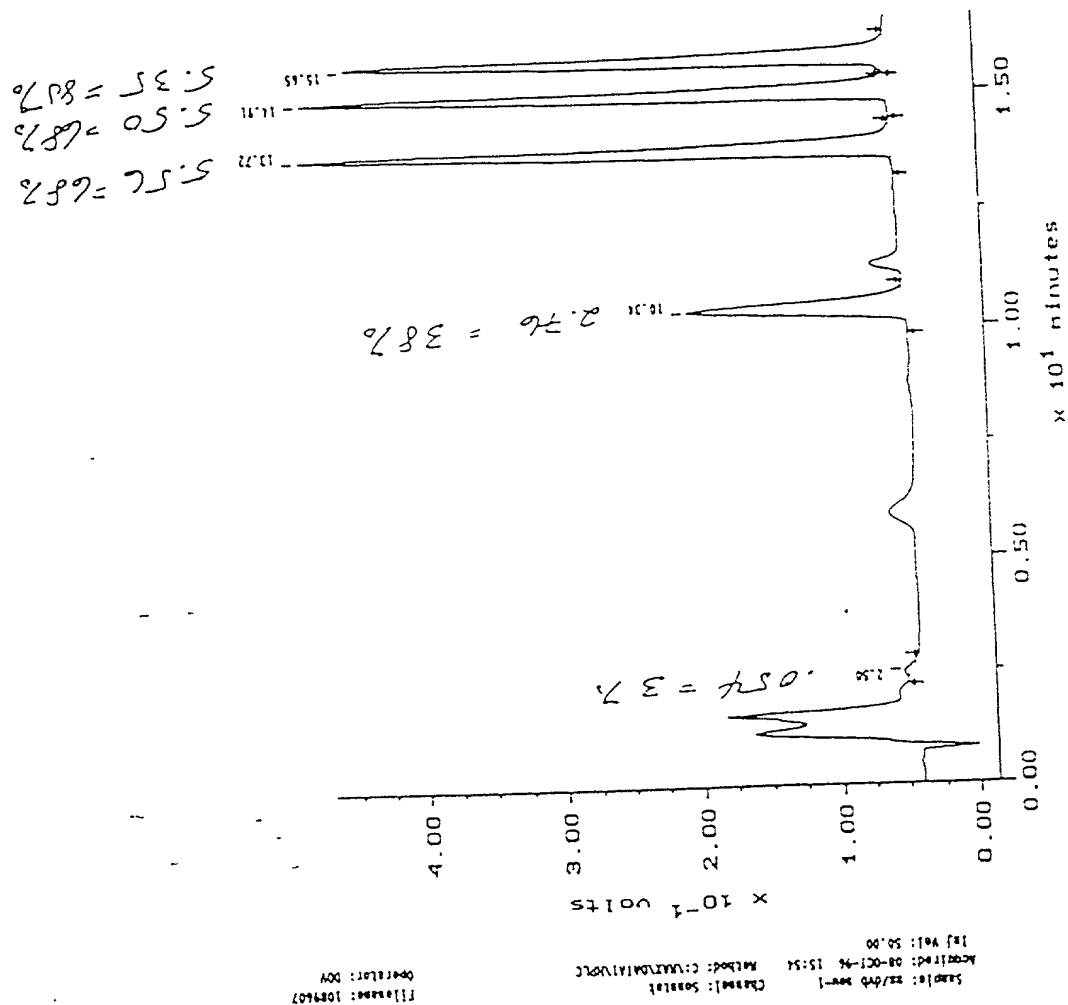
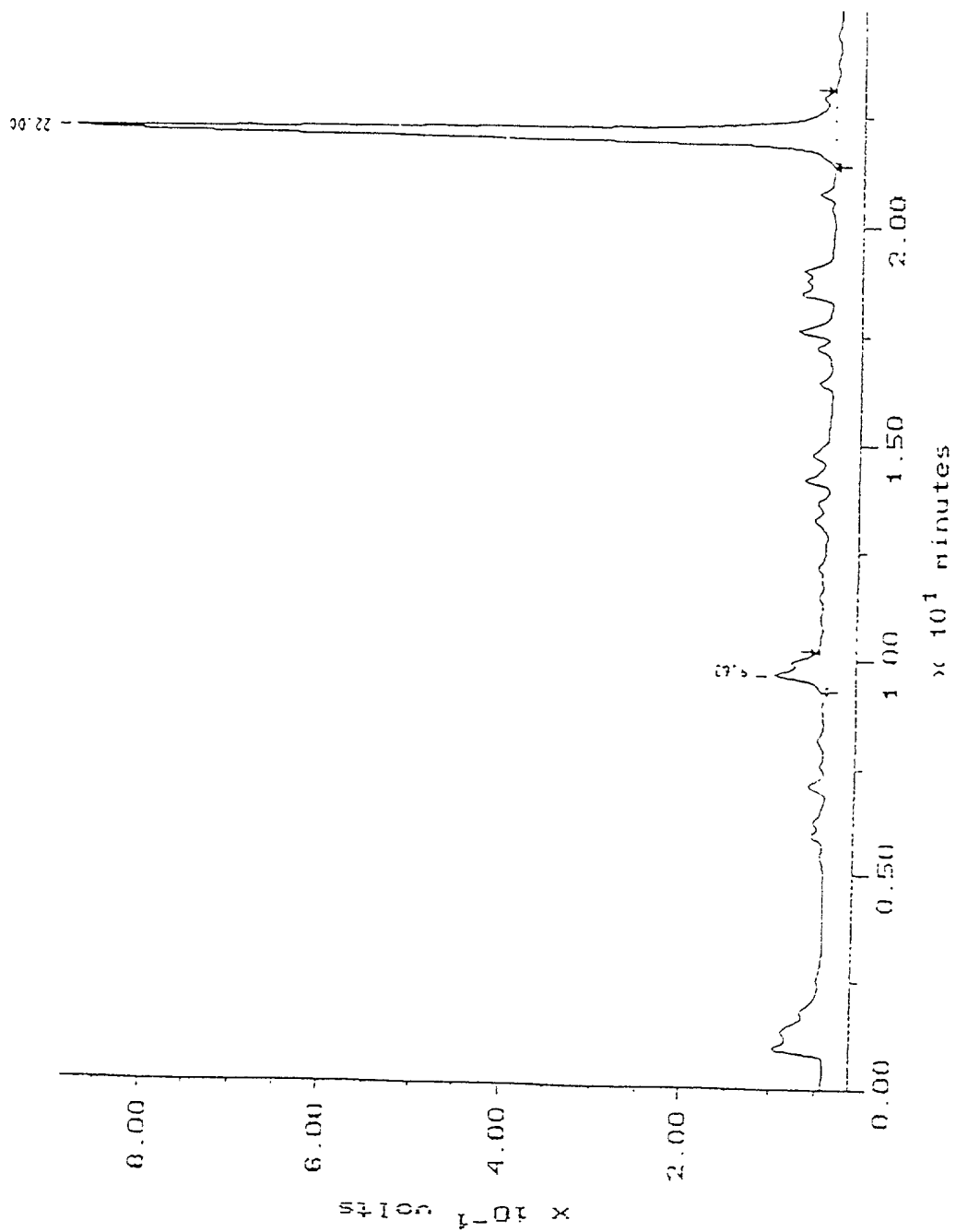




Fig. 9

504000 ETHANE

Cytochrome c



Parameter	Value	Parameter	Value
Temperature	25.0	Temperature	25.0
Pressure	1.0	Pressure	1.0
Humidity	50.0	Humidity	50.0
Wind speed	0.0	Wind speed	0.0
Wind direction	0.0	Wind direction	0.0
Cloud cover	0.0	Cloud cover	0.0
Visibility	10.0	Visibility	10.0
Relative humidity	50.0	Relative humidity	50.0
Dew point	15.0	Dew point	15.0
Sea level pressure	1013.25	Sea level pressure	1013.25
Altitude	0.0	Altitude	0.0
Latitude	0.0	Latitude	0.0
Longitude	0.0	Longitude	0.0
Time of day	0.0	Time of day	0.0
Month	0.0	Month	0.0
Year	0.0	Year	0.0
Day of year	0.0	Day of year	0.0
Week of year	0.0	Week of year	0.0
Hour of day	0.0	Hour of day	0.0
Minute of hour	0.0	Minute of hour	0.0
Second of minute	0.0	Second of minute	0.0
Millisecond of second	0.0	Millisecond of second	0.0
Microsecond of millisecond	0.0	Microsecond of millisecond	0.0
Nanosecond of microsecond	0.0	Nanosecond of microsecond	0.0
Picosecond of nanosecond	0.0	Picosecond of nanosecond	0.0
Femtosecond of picosecond	0.0	Femtosecond of picosecond	0.0
Attosecond of femtosecond	0.0	Attosecond of femtosecond	0.0
Zettosecond of attosecond	0.0	Zettosecond of attosecond	0.0
Yottasecond of zettasecond	0.0	Yottasecond of zettasecond	0.0
Planck time	5.39e-44	Planck time	5.39e-44
Planck length	1.616e-35	Planck length	1.616e-35
Planck mass	2.176e-8	Planck mass	2.176e-8
Planck charge	1.875e-18	Planck charge	1.875e-18
Planck temperature	1.416e+32	Planck temperature	1.416e+32
Planck energy	1.956e-9	Planck energy	1.956e-9
Planck force	1.21e+44	Planck force	1.21e+44
Planck power	3.75e+52	Planck power	3.75e+52
Planck area	2.61e-70	Planck area	2.61e-70
Planck volume	4.22e-107	Planck volume	4.22e-107
Planck density	5.15e+96	Planck density	5.15e+96
Planck pressure	4.79e+113	Planck pressure	4.79e+113
Planck stress	4.79e+113	Planck stress	4.79e+113
Planck energy density	4.79e+113	Planck energy density	4.79e+113
Planck energy flux	4.79e+113	Planck energy flux	4.79e+113
Planck energy flux density	4.79e+113	Planck energy flux density	4.79e+113
Planck energy flux density per unit area	4.79e+113	Planck energy flux density per unit area	4.79e+113
Planck energy flux density per unit volume	4.79e+113	Planck energy flux density per unit volume	4.79e+113
Planck energy flux density per unit mass	4.79e+113	Planck energy flux density per unit mass	4.79e+113
Planck energy flux density per unit length	4.79e+113	Planck energy flux density per unit length	4.79e+113
Planck energy flux density per unit time	4.79e+113	Planck energy flux density per unit time	4.79e+113
Planck energy flux density per unit frequency	4.79e+113	Planck energy flux density per unit frequency	4.79e+113
Planck energy flux density per unit wavelength	4.79e+113	Planck energy flux density per unit wavelength	4.79e+113
Planck energy flux density per unit wave number	4.79e+113	Planck energy flux density per unit wave number	4.79e+113
Planck energy flux density per unit angular frequency	4.79e+113	Planck energy flux density per unit angular frequency	4.79e+113
Planck energy flux density per unit angular wave number	4.79e+113	Planck energy flux density per unit angular wave number	4.79e+113
Planck energy flux density per unit angular frequency per unit angular wave number	4.79e+113	Planck energy flux density per unit angular frequency per unit angular wave number	4.79e+113
Planck energy flux density per unit angular frequency per unit angular wave number per unit area	4.79e+113	Planck energy flux density per unit angular frequency per unit angular wave number per unit area	4.79e+113
Planck energy flux density per unit angular frequency per unit angular wave number per unit volume	4.79e+113	Planck energy flux density per unit angular frequency per unit angular wave number per unit volume	4.79e+113
Planck energy flux density per unit angular frequency per unit angular wave number per unit mass	4.79e+113	Planck energy flux density per unit angular frequency per unit angular wave number per unit mass	4.79e+113
Planck energy flux density per unit angular frequency per unit angular wave number per unit length	4.79e+113	Planck energy flux density per unit angular frequency per unit angular wave number per unit length	4.79e+113
Planck energy flux density per unit angular frequency per unit angular wave number per unit time	4.79e+113	Planck energy flux density per unit angular frequency per unit angular wave number per unit time	4.79e+113
Planck energy flux density per unit angular frequency per unit angular wave number per unit frequency	4.79e+113	Planck energy flux density per unit angular frequency per unit angular wave number per unit frequency	4.79e+113
Planck energy flux density per unit angular frequency per unit angular wave number per unit wavelength	4.79e+113	Planck energy flux density per unit angular frequency per unit angular wave number per unit wavelength	4.79e+113
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Planck energy flux density per unit angular frequency per unit angular wave number per unit angular frequency per unit angular wave number	4.79e+113	Planck energy flux density per unit angular frequency per unit angular wave number per unit angular frequency per unit angular wave number	4.79e+113
Planck energy flux density per unit angular frequency per unit angular wave number per unit angular frequency per unit angular wave number per unit area	4.79e+113	Planck energy flux density per unit angular frequency per unit angular wave number per unit angular frequency per unit angular wave number per unit area	4.79e+113
Planck energy flux density per unit angular frequency per unit angular wave number per unit angular frequency per unit angular wave number per unit volume	4.79e+113	Planck energy flux density per unit angular frequency per unit angular wave number per unit angular frequency per unit angular wave number per unit volume	4.79e+113
Planck energy flux density per unit angular frequency per unit angular wave number per			

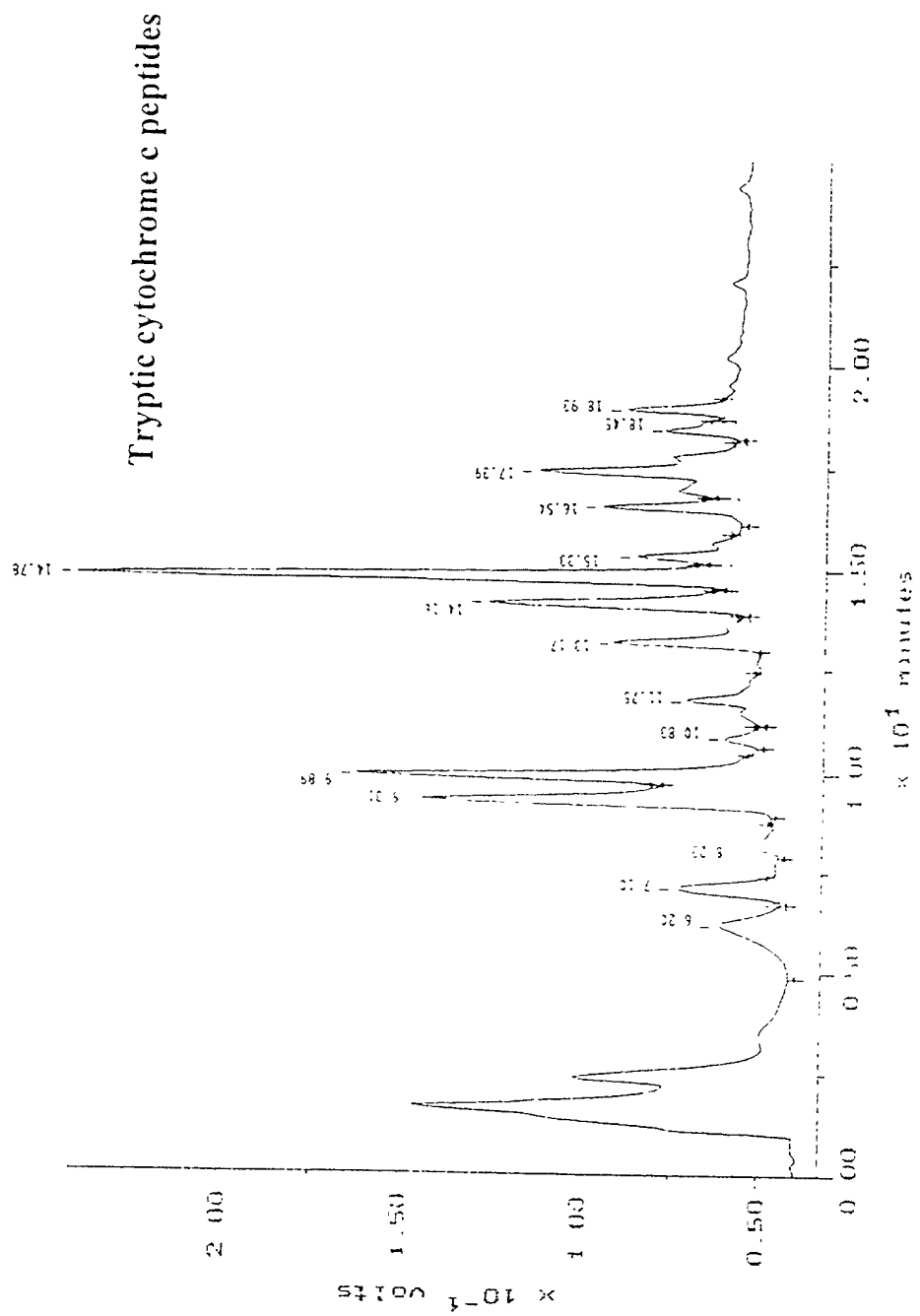


Fig. 11

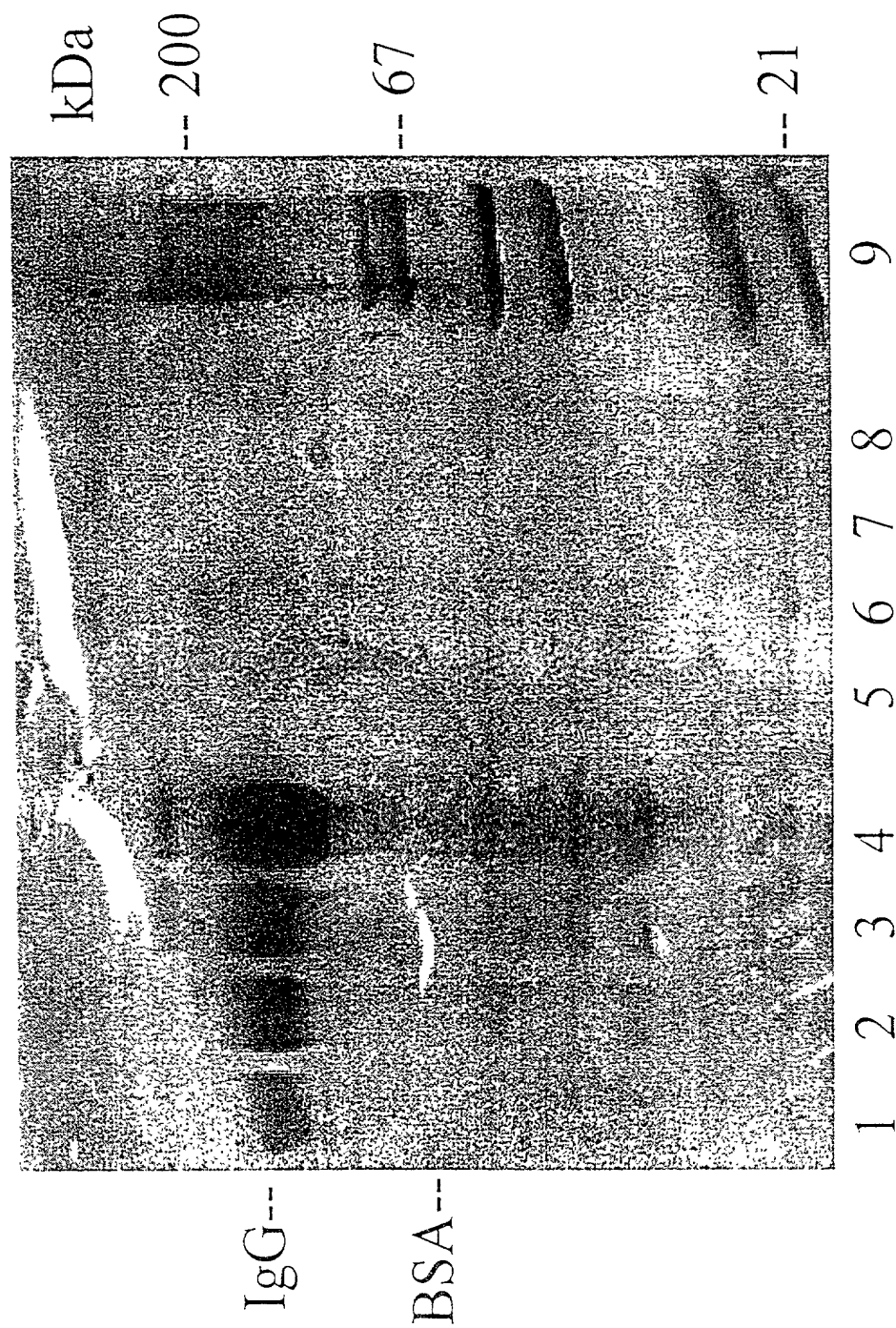
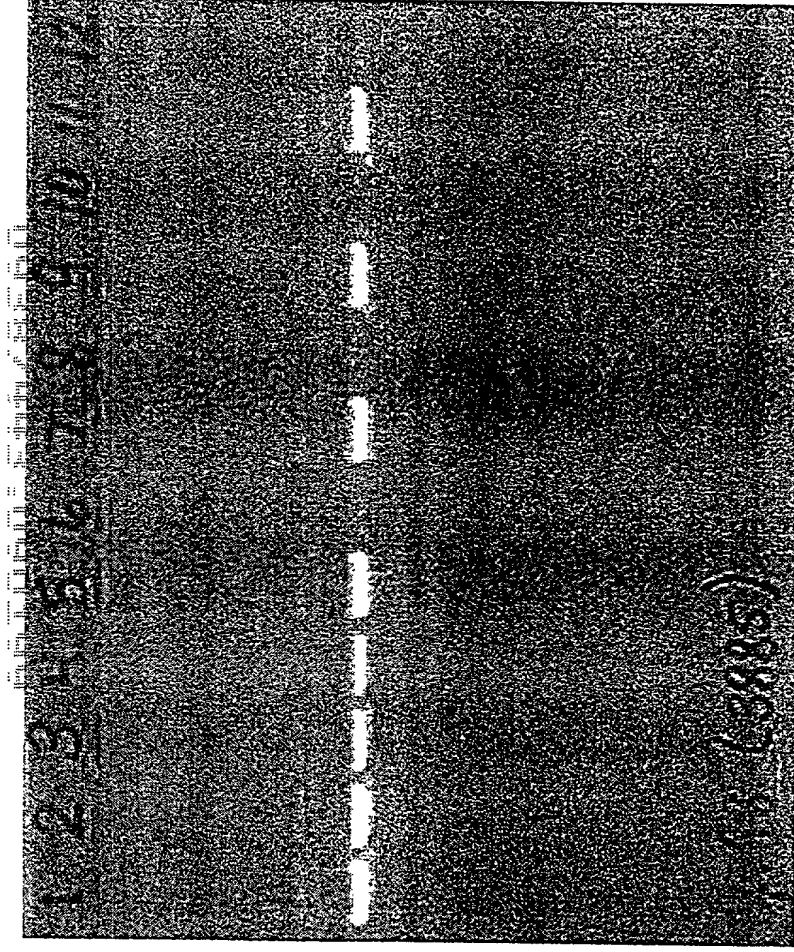


FIG. 12



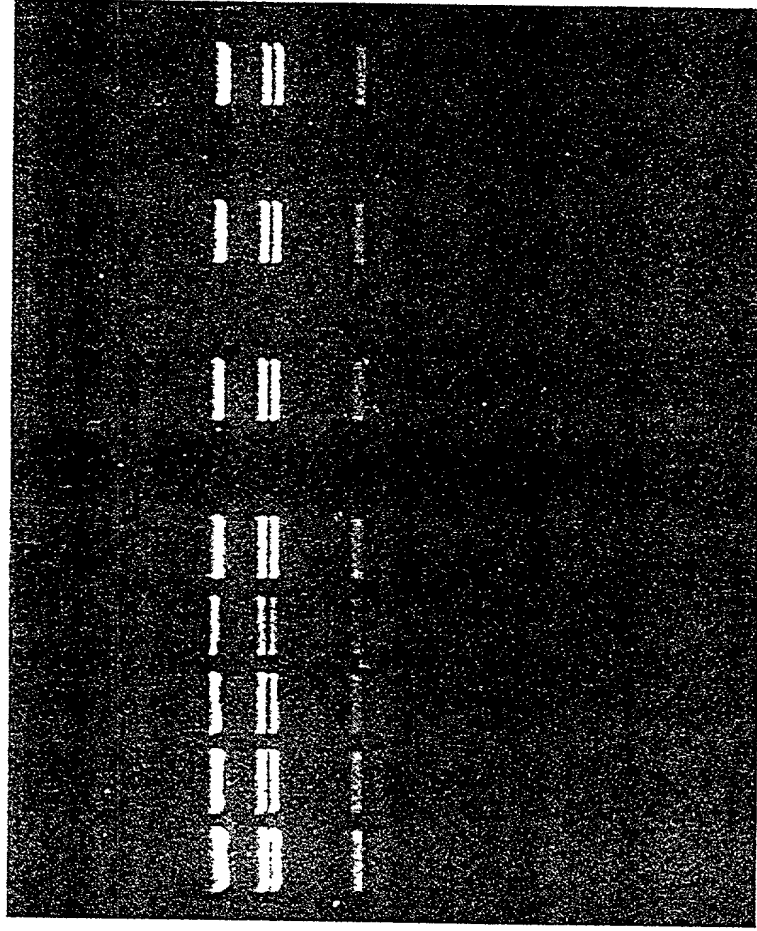
lane

1. 100 % recovery equivalent
2. 75 % recovery equivalent
3. 50 % recovery equivalent
4. 25 % recovery equivalent
5. Tip #1 eluate
6. Tip #1 flow through

lane

7. Tip #2 eluate
8. Tip #2 flow through
9. Tip #3 eluate
10. Tip #3 flow through
11. Tip #4 eluate
12. Tip #4 flow through

FIG. 13



1 2 3 4 5 6 7 8 9 10 11 12

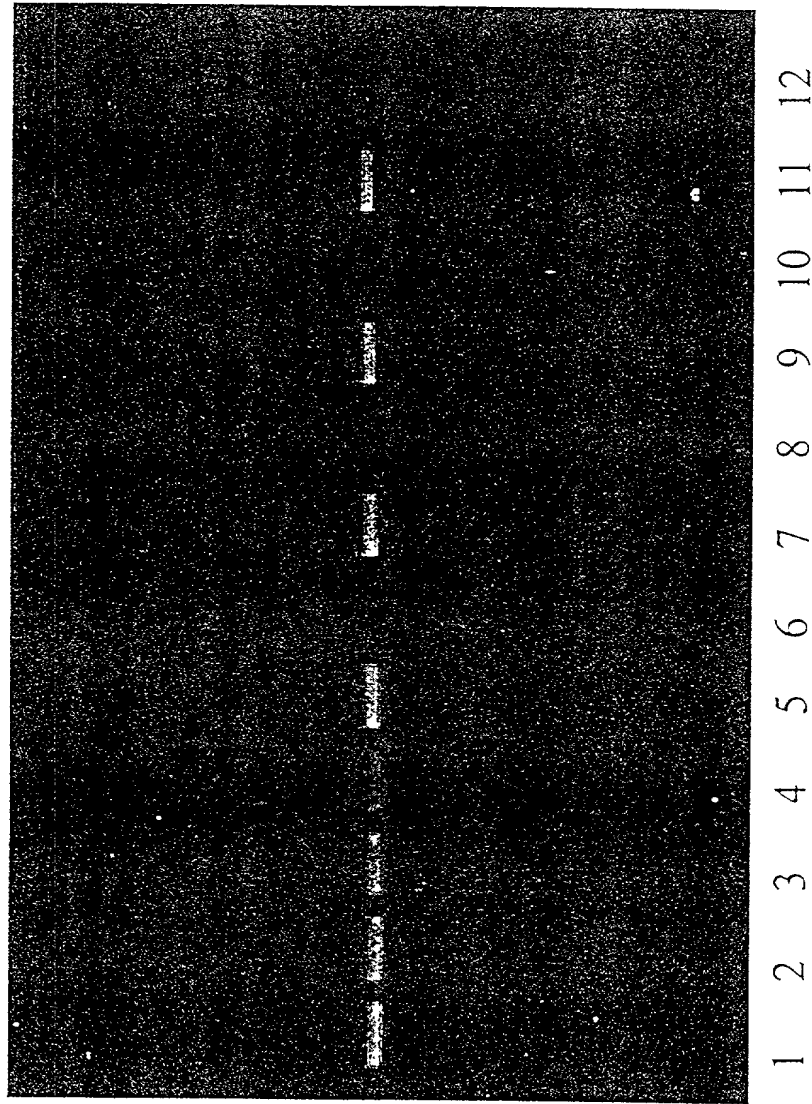
lane

1. 100 % recovery equivalent
2. 75 % recovery equivalent
3. 50 % recovery equivalent
4. 25 % recovery equivalent
5. Tip #1 eluate
6. Tip #1 flow through

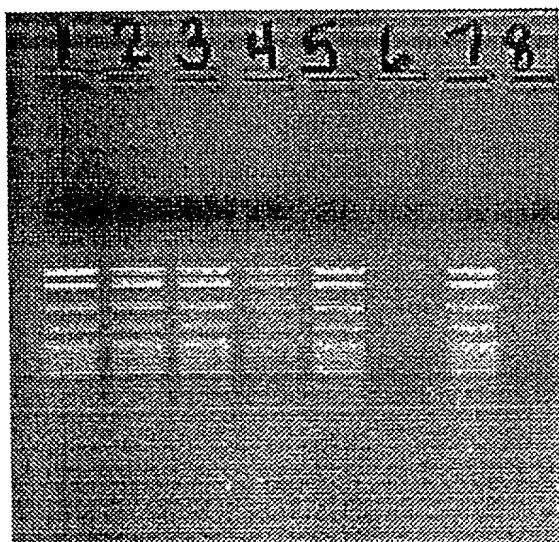
lane

7. Tip #2 eluate
8. Tip #2 flow through
9. Tip #3 eluate
10. Tip #3 flow through
11. Tip #4 eluate
12. Tip #4 flow through

FIG. 14



<u>lane</u>	<u>lane</u>
1. 100 % recovery equivalent	7. Tip #2 eluate
2. 75 % recovery equivalent	8. Tip #2 flow through
3. 50 % recovery equivalent	9. Tip #3 eluate
4. 25 % recovery equivalent	10. Tip #3 flow through
5. Tip #1 eluate	11. Tip #4 eluate
6. Tip #1 flow through	12. Tip #4 flow through



LANE

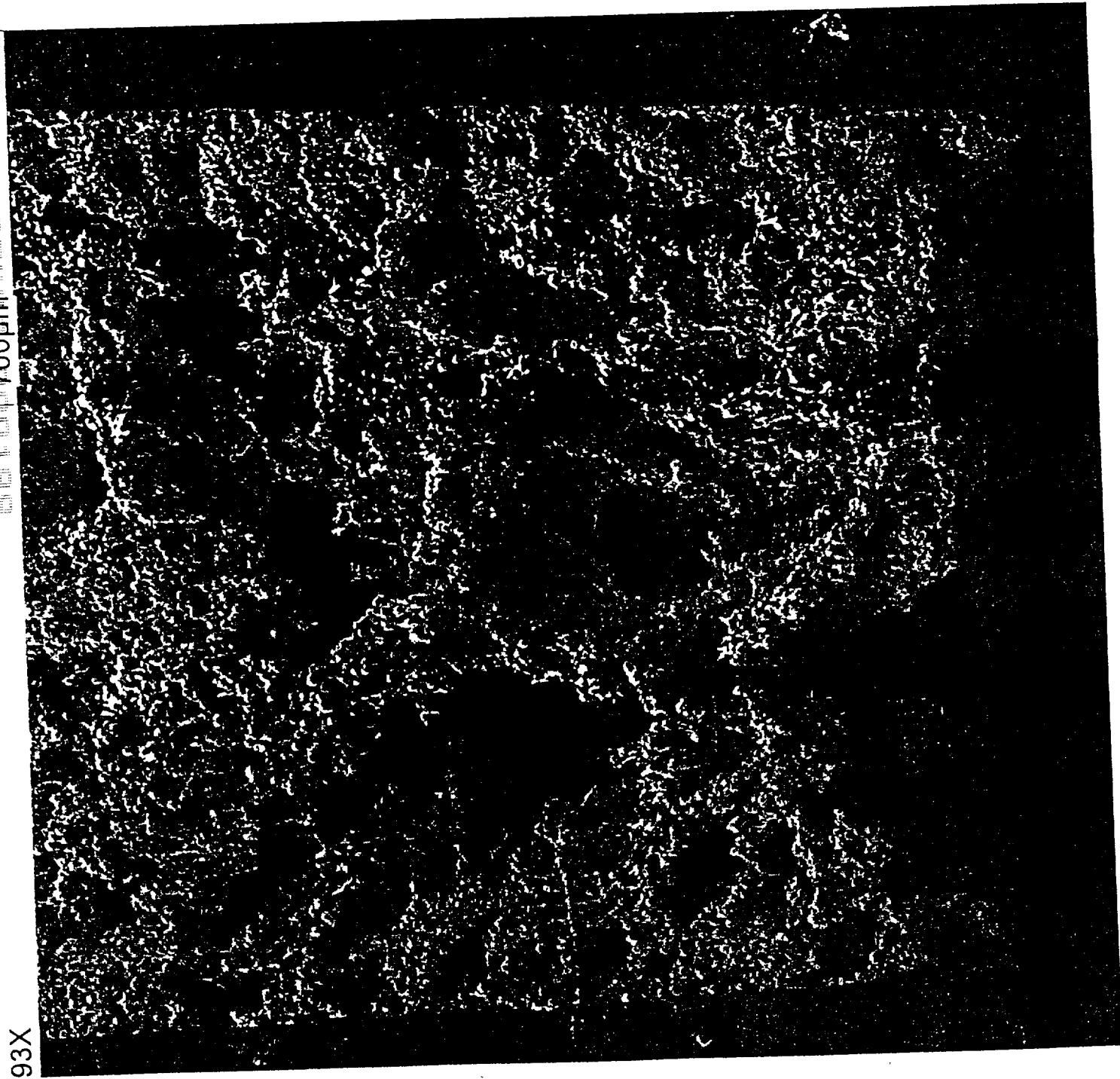
1. 100% recovery equivalent
2. 75% recovery equivalent
3. 50% recovery equivalent
4. 25% recovery equivalent
5. Tip #1 eluate
6. Tip #1 flow through
7. Tip #2 eluate
8. Tip #2 flow through

FIG. 15

93X

500 1000 700µm

FIG. 16A





66F020-27046600

720μm

91X

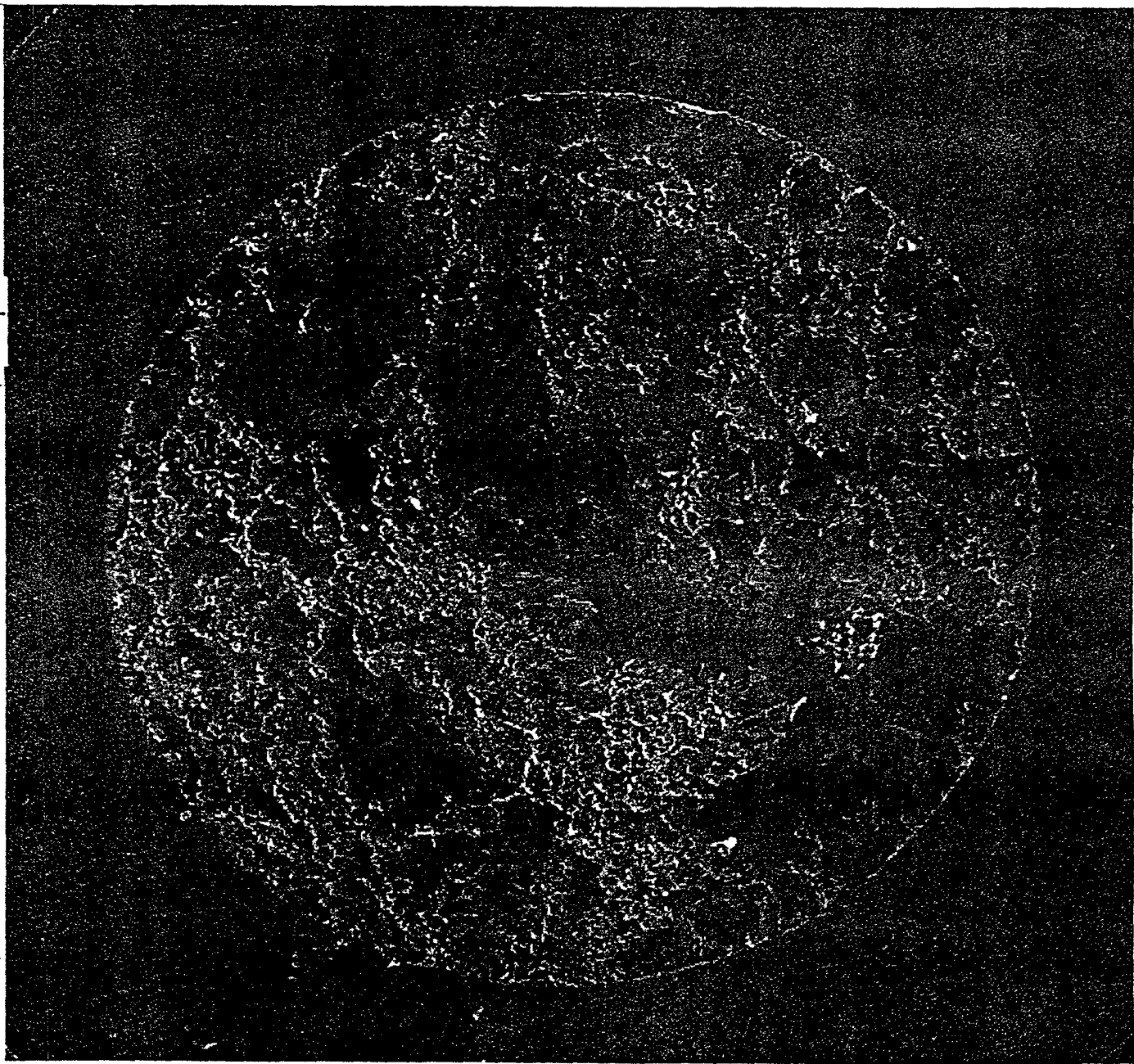


FIG. 16B

380X

170µm

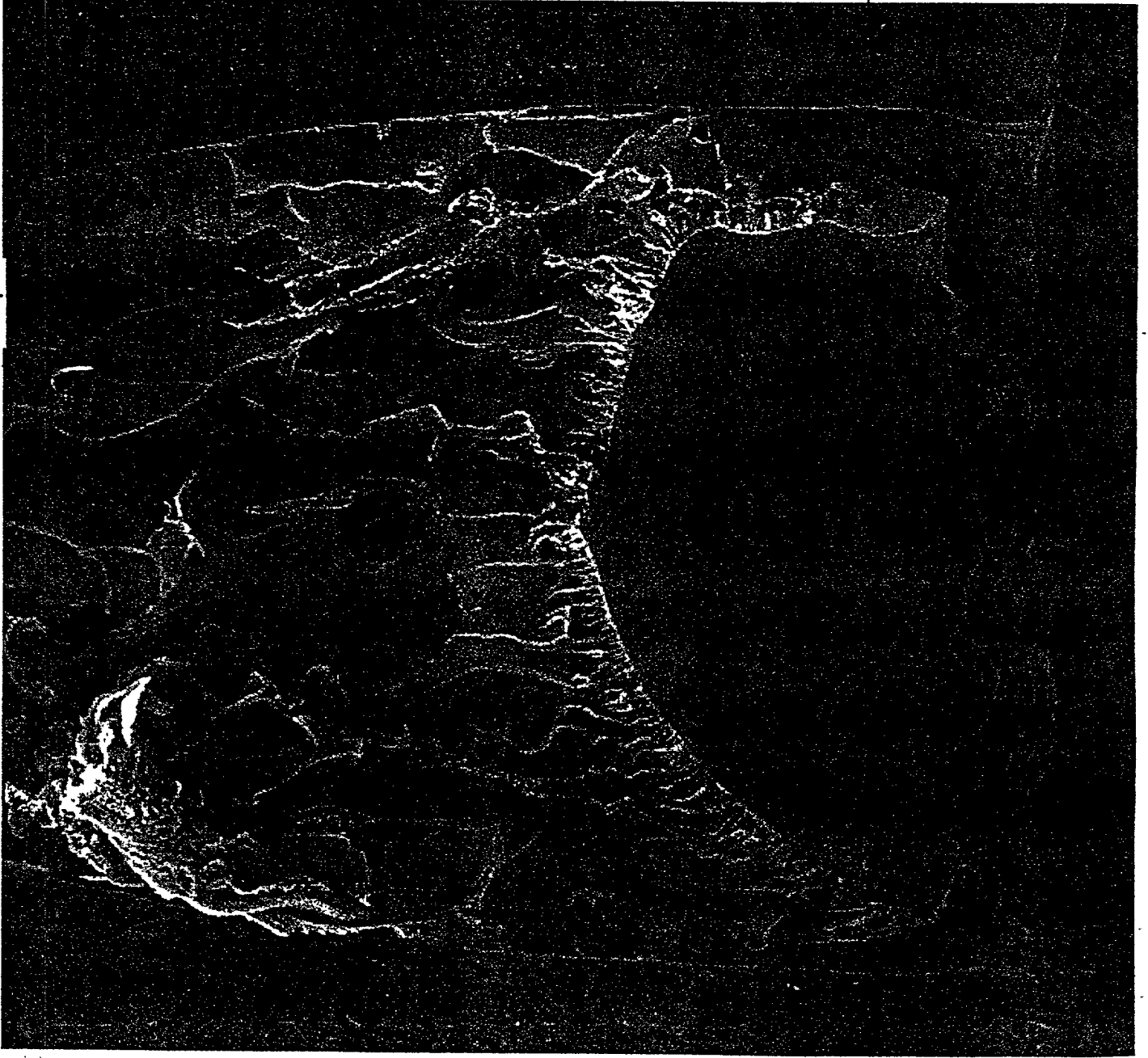


FIG. 17

## DECLARATION FOR PATENT APPLICATION

Docket Number (Optional)

550P002

As a below named inventor, I hereby declare that:

My residence, post office address and citizenship are as stated below next to my name.

I believe I am the original, first and sole inventor (if only one name is listed below) or an original, first and joint inventor (if plural names are listed below) of the subject matter which is claimed and for which a patent is sought on the invention entitled  
CAST MEMBRANE STRUCTURES FOR SAMPLE PREPARATION, the specification of which

is attached hereto unless the following box is checked:

☒ was filed on January 15, 1998 as United States Application Number or PCT International Application Number 09/007,320 and was amended on \_\_\_\_\_ (if applicable).

I hereby state that I have reviewed and understand the contents of the above identified specification, including the claims, as amended by any amendment referred to above.

I acknowledge the duty to disclose information which is material to patentability as defined in 37 CFR § 1.56.

I hereby claim foreign priority benefits under 35 U.S.C. § 119(a)-(d) or § 365(b) of any foreign application(s) for patent or inventor's certificate, or § 365(a) of any PCT International application which designated at least one country other than the United States, listed below and have also identified below, by checking the box, any foreign application for patent or inventor's certificate, or PCT International application having a filing date before that of the application on which priority is claimed.

Prior Foreign Application(s)

Priority Not Claimed

_____ (Number)	_____ (Country)	_____ (Day/Month/Year Filed)	<input type="checkbox"/>
_____ (Number)	_____ (Country)	_____ (Day/Month/Year Filed)	<input type="checkbox"/>

I hereby claim the benefit under 35 U.S.C. § 119(e) of any United States provisional application(s) listed below

<u>60/038,909</u> (Application Number)	<u>February 26, 1997</u> (Filing Date)
_____ (Application Number)	_____ (Filing Date)

I hereby claim the benefit under 35 U.S.C. § 120 of any United States application(s), or § 365(c) of any PCT International application designating the United States, listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in the prior United States or PCT International application in the manner provided by the first paragraph of 35 U.S.C. § 112.

I acknowledge the duty to disclose information which is material to patentability as defined in 37 CFR § 1.56 which became available between the filing date of the prior application and the national or PCT International filing date of this application.

_____ (Application Number)	_____ (Filing Date)	_____ (Status -- patented, pending, abandoned)
_____ (Application Number)	_____ (Filing Date)	_____ (Status -- patented, pending, abandoned)

I hereby appoint the following attorney(s) and/or agent(s) to prosecute this application and to transact all business in the Patent and Trademark Office connected therewith:

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I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issued thereon.

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Second inventor's signature Donald G. Sheer Date 3/5/98  
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☒ Additional inventors are being named on separately numbered sheets attached hereto.

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